

# Hordenine

<b>Other names:</b>	Phenol, 4-[2-(dimethylamino)ethyl]- Phenol, p-[2-(dimethylamino)ethyl]- p-[2-(Dimethylamino)ethyl]phenol p-Hydroxy-N,N-dimethylphenethylamine Anhalin Anhaline Eremursine Hordenin N,N-Dimethyl-p-hydroxyphenethylamine N,N-Dimethyltyramine Ordenina Ordenine Peyocactine N,N-Dimethyl-2-(4-hydroxyphenyl)ethylamine Cactine N,N-Dimethyl-4-hydroxy-«beta»-phenethylamine 4-(2-dimethylaminoethyl)phenol
<b>Inchi:</b>	InChI=1S/C10H15NO/c1-11(2)8-7-9-3-5-10(12)6-4-9/h3-6,12H,7-8H2,1-2H3
<b>InchiKey:</b>	KUBCEEMXQZUPDQ-UHFFFAOYSA-N
<b>Formula:</b>	C10H15NO
<b>SMILES:</b>	CN(C)CCc1ccc(O)cc1
<b>Mol. weight [g/mol]:</b>	165.23
<b>CAS:</b>	539-15-1

## Physical Properties

Property code	Value	Unit	Source
gf	101.89	kJ/mol	Joback Method
hf	-122.98	kJ/mol	Joback Method
hfus	24.50	kJ/mol	Joback Method
hvap	55.19	kJ/mol	Joback Method
log10ws	-1.23		Crippen Method
logp	1.496		Crippen Method
mcvol	143.850	ml/mol	McGowan Method
pc	3460.21	kPa	Joback Method
rinpol	1485.00		NIST Webbook
rinpol	1490.00		NIST Webbook
rinpol	1490.00		NIST Webbook

rmpol	1495.00		NIST Webbook
rmpol	1495.00		NIST Webbook
rmpol	1485.00		NIST Webbook
tb	547.94	K	Joback Method
tc	763.06	K	Joback Method
tf	373.07	K	Joback Method
vc	0.471	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	345.95	J/mol×K	547.94	Joback Method
cpg	360.43	J/mol×K	583.79	Joback Method
cpg	373.90	J/mol×K	619.65	Joback Method
cpg	386.47	J/mol×K	655.50	Joback Method
cpg	398.20	J/mol×K	691.35	Joback Method
cpg	409.19	J/mol×K	727.20	Joback Method
cpg	419.51	J/mol×K	763.06	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C539151&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C539151&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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