

# «gamma»-Aminobutyric acid, N-isobutoxycarbonyl-, heptadecyl ester

**Other names:** .gama.-Aminobutyric acid, N-isobutoxycarbonyl-, heptadecyl ester

**Inchi:** InChI=1S/C26H51NO4/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-22-30-25(28)20-19-2

**InchiKey:** RGQFMKNSSWBX-UHFFFAOYSA-N

**Formula:** C26H51NO4

**SMILES:** CCCCCCCCCCCCCCCCCOC(=O)CCNC(=O)OCC(C)C

**Mol. weight [g/mol]:** 441.69

## Physical Properties

Property code	Value	Unit	Source
gf	-212.85	kJ/mol	Joback Method
hf	-1021.38	kJ/mol	Joback Method
hfus	70.25	kJ/mol	Joback Method
hvap	97.83	kJ/mol	Joback Method
log10ws	-8.36		Crippen Method
logp	7.563		Crippen Method
mcvol	402.060	ml/mol	McGowan Method
pc	771.18	kPa	Joback Method
tb	996.59	K	Joback Method
tc	1233.25	K	Joback Method
tf	564.76	K	Joback Method
vc	1.569	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1411.28	J/molxK	996.59	Joback Method
cpg	1432.12	J/molxK	1036.03	Joback Method
cpg	1451.03	J/molxK	1075.48	Joback Method
cpg	1468.07	J/molxK	1114.92	Joback Method
cpg	1483.30	J/molxK	1154.36	Joback Method
cpg	1496.80	J/molxK	1193.81	Joback Method
cpg	1508.63	J/molxK	1233.25	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U321059&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U321059&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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