

(2R,3R)-(-)-2-Benzyloxy-1,3,4-butanetriol, tris(trifluoroacetate)

Inchi: InChI=1S/C17H13F9O7/c18-15(19,20)12(27)31-7-10(30-6-9-4-2-1-3-5-9)11(33-14(29)17
InchiKey: QVSPHGYEGENNA-UHFFFAOYSA-N
Formula: C17H13F9O7
SMILES: O=C(OCC(OCC1OC1)C(COC(=O)C(F)(F)F)OC(=O)C(F)(F)F)C(F)(F)F
Mol. weight [g/mol]: 500.27

Physical Properties

Property code	Value	Unit	Source
gf	-2351.74	kJ/mol	Joback Method
hf	-2826.10	kJ/mol	Joback Method
hfus	41.81	kJ/mol	Joback Method
hvap	73.57	kJ/mol	Joback Method
log10ws	-4.42		Crippen Method
logp	3.257		Crippen Method
mcvol	270.750	ml/mol	McGowan Method
pc	1341.76	kPa	Joback Method
rinpol	1496.10		NIST Webbook
tb	849.19	K	Joback Method
tc	1041.35	K	Joback Method
tf	529.05	K	Joback Method
vc	1.087	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	847.81	J/molxK	849.19	Joback Method
cpg	858.10	J/molxK	881.22	Joback Method
cpg	867.40	J/molxK	913.24	Joback Method
cpg	875.76	J/molxK	945.27	Joback Method
cpg	883.24	J/molxK	977.29	Joback Method
cpg	889.87	J/molxK	1009.32	Joback Method
cpg	895.70	J/molxK	1041.35	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U380215&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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