

# Benzene,2-ethoxy-1,3-dimethyl-

<b>Inchi:</b>	InChI=1S/C10H14O/c1-4-11-10-8(2)6-5-7-9(10)3/h5-7H,4H2,1-3H3
<b>InchiKey:</b>	HCEKYXXNVJXAIF-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	CCOc1c(C)cccc1C
<b>Mol. weight [g/mol]:</b>	150.22
<b>CAS:</b>	26620-08-6

## Physical Properties

Property code	Value	Unit	Source
gf	21.47	kJ/mol	Joback Method
hf	-168.36	kJ/mol	Joback Method
hfus	16.11	kJ/mol	Joback Method
hvap	43.86	kJ/mol	Joback Method
ie	8.49	eV	NIST Webbook
log10ws	-2.96		Crippen Method
logp	2.702		Crippen Method
mcvol	133.870	ml/mol	McGowan Method
pc	2787.66	kPa	Joback Method
rinpol	1308.00		NIST Webbook
tb	487.26	K	Joback Method
tc	692.88	K	Joback Method
tf	276.15	K	Joback Method
vc	0.505	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	282.14	J/molxK	487.26	Joback Method
cpg	296.04	J/molxK	521.53	Joback Method
cpg	309.33	J/molxK	555.80	Joback Method
cpg	322.01	J/molxK	590.07	Joback Method
cpg	334.09	J/molxK	624.34	Joback Method
cpg	345.58	J/molxK	658.61	Joback Method
cpg	356.48	J/molxK	692.88	Joback Method

dvisc	0.0013933	Paxs	276.15	Joback Method
dvisc	0.0008168	Paxs	311.33	Joback Method
dvisc	0.0005336	Paxs	346.52	Joback Method
dvisc	0.0003771	Paxs	381.70	Joback Method
dvisc	0.0002826	Paxs	416.89	Joback Method
dvisc	0.0002215	Paxs	452.07	Joback Method
dvisc	0.0001798	Paxs	487.26	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C26620086&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C26620086&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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