

# Diethylmalonic acid, 2-hexyl octadecyl ester

<b>Inchi:</b>	InChI=1S/C31H60O4/c1-6-10-12-13-14-15-16-17-18-19-20-21-22-23-24-25-27-34-29(32)
<b>InchiKey:</b>	JIGSVBACBMTGPN-UHFFFAOYSA-N
<b>Formula:</b>	C31H60O4
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OC(C)CCCC
<b>Mol. weight [g/mol]:</b>	496.81

## Physical Properties

Property code	Value	Unit	Source
gf	-257.30	kJ/mol	Joback Method
hf	-1186.80	kJ/mol	Joback Method
hfus	70.68	kJ/mol	Joback Method
hvap	101.23	kJ/mol	Joback Method
log10ws	-10.39		Crippen Method
logp	9.719		Crippen Method
mcvol	462.530	ml/mol	McGowan Method
pc	602.21	kPa	Joback Method
rinpol	3083.00		NIST Webbook
tb	1057.59	K	Joback Method
tc	1325.01	K	Joback Method
tf	570.87	K	Joback Method
vc	1.802	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1670.72	J/molxK	1057.59	Joback Method
cpg	1770.52	J/molxK	1280.44	Joback Method
cpg	1754.54	J/molxK	1235.87	Joback Method
cpg	1736.73	J/molxK	1191.30	Joback Method
cpg	1716.93	J/molxK	1146.73	Joback Method
cpg	1694.98	J/molxK	1102.16	Joback Method
cpg	1784.83	J/molxK	1325.01	Joback Method
dvisc	0.0000066	Paxs	1057.59	Joback Method
dvisc	0.0000092	Paxs	976.47	Joback Method

dvisc	0.0000136	Paxs	895.35	Joback Method
dvisc	0.0000218	Paxs	814.23	Joback Method
dvisc	0.0000389	Paxs	733.11	Joback Method
dvisc	0.0000799	Paxs	651.99	Joback Method
dvisc	0.0002015	Paxs	570.87	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370636&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370636&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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