

Ethylmalonic acid

Other names:	Propanedioic acid, ethyl- «alpha»-Carboxybutyric acid Malonic acid, ethyl- 1,1-Propanedicarboxylic acid
Inchi:	InChI=1S/C5H8O4/c1-2-3(4(6)7)5(8)9/h3H,2H2,1H3,(H,6,7)(H,8,9)
InchiKey:	UKFXDFUAPNAMPJ-UHFFFAOYSA-N
Formula:	C5H8O4
SMILES:	CCC(C(=O)O)C(=O)O
Mol. weight [g/mol]:	132.11
CAS:	601-75-2

Physical Properties

Property code	Value	Unit	Source
chs	-2172.20	kJ/mol	NIST Webbook
gf	-542.70	kJ/mol	Joback Method
hf	-849.80 ± 1.10	kJ/mol	NIST Webbook
hfs	-955.30 ± 1.00	kJ/mol	NIST Webbook
hfus	16.56	kJ/mol	Joback Method
hsub	105.50	kJ/mol	NIST Webbook
hsub	105.50 ± 0.50	kJ/mol	NIST Webbook
hsub	112.80 ± 2.20	kJ/mol	NIST Webbook
hvap	73.19	kJ/mol	Joback Method
log10ws	0.13		Crippen Method
logp	0.182		Crippen Method
mcvol	96.190	ml/mol	McGowan Method
pc	5251.00	kPa	Joback Method
tb	605.46	K	Joback Method
tc	783.09	K	Joback Method
tf	384.00 ± 3.00	K	NIST Webbook
vc	0.359	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	232.92	J/molxK	605.46	Joback Method
cpg	261.32	J/molxK	753.49	Joback Method
cpg	256.24	J/molxK	723.88	Joback Method
cpg	250.87	J/molxK	694.28	Joback Method
cpg	245.20	J/molxK	664.67	Joback Method
cpg	239.21	J/molxK	635.07	Joback Method
cpg	266.11	J/molxK	783.09	Joback Method
dvisc	0.0000446	Paxs	605.46	Joback Method
dvisc	0.0000803	Paxs	563.32	Joback Method
dvisc	0.0001588	Paxs	521.18	Joback Method
dvisc	0.0003540	Paxs	479.04	Joback Method
dvisc	0.0009213	Paxs	436.89	Joback Method
dvisc	0.0029414	Paxs	394.75	Joback Method
dvisc	0.0123933	Paxs	352.61	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C601752&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

chs:	Standard solid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfs:	Solid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hsub:	Enthalpy of sublimation at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure

tb: Normal Boiling Point Temperature
tc: Critical Temperature
tf: Normal melting (fusion) point
vc: Critical Volume

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