

# S-Methyl propanethioate

<b>Other names:</b>	S-Methyl thiopropanoate S-Methyl thiopropionate S-Methyl methylthiopropionate
<b>Inchi:</b>	InChI=1S/C4H8OS/c1-3-4(5)6-2/h3H2,1-2H3
<b>InchiKey:</b>	AIILTVHCLAEMDA-UHFFFAOYSA-N
<b>Formula:</b>	C4H8OS
<b>SMILES:</b>	CCC(=O)SC
<b>Mol. weight [g/mol]:</b>	104.17
<b>CAS:</b>	5925-75-7

## Physical Properties

Property code	Value	Unit	Source
gf	-113.00	kJ/mol	Joback Method
hf	-196.60	kJ/mol	Joback Method
hfus	11.84	kJ/mol	Joback Method
hvap	38.06	kJ/mol	Joback Method
log10ws	-1.16		Crippen Method
logp	1.286		Crippen Method
mcvol	85.140	ml/mol	McGowan Method
pc	4316.89	kPa	Joback Method
rinpol	785.00		NIST Webbook
rinpol	785.00		NIST Webbook
rinpol	785.00		NIST Webbook
rinpol	785.00		NIST Webbook
ripol	1130.00		NIST Webbook
ripol	1131.00		NIST Webbook
ripol	1130.00		NIST Webbook
ripol	1131.00		NIST Webbook
tb	413.57	K	Joback Method
tc	618.94	K	Joback Method
tf	219.17	K	Joback Method
vc	0.320	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	147.97	J/mol×K	413.57	Joback Method
cpg	155.97	J/mol×K	447.80	Joback Method
cpg	163.66	J/mol×K	482.03	Joback Method
cpg	171.03	J/mol×K	516.25	Joback Method
cpg	178.09	J/mol×K	550.48	Joback Method
cpg	184.83	J/mol×K	584.71	Joback Method
cpg	191.25	J/mol×K	618.94	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5925757&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5925757&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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