

# 2-Butenenitrile

<b>Other names:</b>	1-Cyano-1-propylene 1-Cyanopropene 1-Propenyl cyanide Croton acid nitrile Crotonic nitrile Crotonique nitrile Crotonitrile Crotonnitrile Crotononitrile NSC 165574 cis or trans-crotononitrile trans or cis-crotononitrile «beta»-Methylacrylonitrile Â«betaÂ»-Methylacrylonitrile
<b>Inchi:</b>	InChI=1S/C4H5N/c1-2-3-4-5/h2-3H,1H3
<b>InchiKey:</b>	NKKMVIVFRUYPLQ-UHFFFAOYSA-N
<b>Formula:</b>	C4H5N
<b>SMILES:</b>	CC=CC#N
<b>Mol. weight [g/mol]:</b>	67.09
<b>CAS:</b>	4786-20-3

## Physical Properties

Property code	Value	Unit	Source
gf	196.20	kJ/mol	Joback Method
hf	156.21	kJ/mol	Joback Method
hfus	7.82	kJ/mol	Joback Method
hvap	34.93	kJ/mol	Joback Method
log10ws	-1.22		Crippen Method
logp	1.086		Crippen Method
mcvol	64.300	ml/mol	McGowan Method
pc	4031.24	kPa	Joback Method
rinpola	664.00		NIST Webbook
rinpola	697.00		NIST Webbook
rinpola	661.00		NIST Webbook
rinpola	624.00		NIST Webbook
rinpola	661.00		NIST Webbook
rinpola	664.00		NIST Webbook

rinpol	619.80		NIST Webbook
rinpol	647.20		NIST Webbook
rinpol	624.00		NIST Webbook
rinpol	624.00		NIST Webbook
rinpol	664.00		NIST Webbook
rinpol	664.00		NIST Webbook
ripol	1162.00		NIST Webbook
ripol	1162.00		NIST Webbook
ripol	1162.00		NIST Webbook
tb	393.70	K	NIST Webbook
tc	599.62	K	Joback Method
tf	194.75	K	Joback Method
vc	0.266	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	102.51	J/mol×K	397.16	Joback Method
cpg	108.39	J/mol×K	430.90	Joback Method
cpg	113.94	J/mol×K	464.65	Joback Method
cpg	119.18	J/mol×K	498.39	Joback Method
cpg	124.11	J/mol×K	532.13	Joback Method
cpg	128.76	J/mol×K	565.87	Joback Method
cpg	133.16	J/mol×K	599.62	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.46058e+01
Coeff. B	-3.42685e+03
Coeff. C	-5.05850e+01
Temperature range (K), min.	289.92
Temperature range (K), max.	419.29

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4786203&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4786203&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pv<sub>ap</sub>:</b>	Vapor pressure
<b>ri<sub>npol</sub>:</b>	Non-polar retention indices
<b>ri<sub>pol</sub>:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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