

# Phenol, 2,3,5,6-tetramethyl-

<b>Other names:</b>	2,3,5,6-Tetramethylphenol Durenol Phenol, tetramethyl-
<b>Inchi:</b>	InChI=1S/C10H14O/c1-6-5-7(2)9(4)10(11)8(6)3/h5,11H,1-4H3
<b>InchiKey:</b>	KLAQSPUVCDBEGF-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	<chem>Cc1cc(C)c(C)c(O)c1C</chem>
<b>Mol. weight [g/mol]:</b>	150.22
<b>CAS:</b>	527-35-5

## Physical Properties

Property code	Value	Unit	Source
gf	-37.78	kJ/mol	Joback Method
hf	-224.92	kJ/mol	Joback Method
hfus	20.31	kJ/mol	Joback Method
hvap	55.13	kJ/mol	Joback Method
log10ws	-2.92		Crippen Method
logp	2.626		Crippen Method
mcvol	133.870	ml/mol	McGowan Method
pc	3291.59	kPa	Joback Method
rinpol	1287.00		NIST Webbook
rinpol	1354.00		NIST Webbook
rinpol	231.30		NIST Webbook
rinpol	1361.00		NIST Webbook
rinpol	232.59		NIST Webbook
rinpol	1319.00		NIST Webbook
rinpol	1319.00		NIST Webbook
rinpol	231.30		NIST Webbook
rinpol	1354.00		NIST Webbook
tb	520.70 ± 4.00	K	NIST Webbook
tb	522.00 ± 3.00	K	NIST Webbook
tb	520.70 ± 4.00	K	NIST Webbook
tc	772.15	K	Joback Method
tf	392.00 ± 2.00	K	NIST Webbook
tf	390.00 ± 2.00	K	NIST Webbook
tf	390.00 ± 2.00	K	NIST Webbook
tf	392.00 ± 2.00	K	NIST Webbook

tf	389.00 ± 3.00	K	NIST Webbook
vc	0.454	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	310.51	J/mol×K	550.44	Joback Method
cpg	323.33	J/mol×K	587.39	Joback Method
cpg	335.42	J/mol×K	624.34	Joback Method
cpg	346.83	J/mol×K	661.30	Joback Method
cpg	357.63	J/mol×K	698.25	Joback Method
cpg	367.86	J/mol×K	735.20	Joback Method
cpg	377.60	J/mol×K	772.15	Joback Method
dvisc	0.0010235	Paxs	378.16	Joback Method
dvisc	0.0005185	Paxs	406.87	Joback Method
dvisc	0.0002873	Paxs	435.59	Joback Method
dvisc	0.0001712	Paxs	464.30	Joback Method
dvisc	0.0001084	Paxs	493.01	Joback Method
dvisc	0.0000722	Paxs	521.73	Joback Method
dvisc	0.0000501	Paxs	550.44	Joback Method
hvapt	51.20	kJ/mol	451.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54956e+01
Coeff. B	-4.72801e+03
Coeff. C	-8.60300e+01
Temperature range (K), min.	390.15
Temperature range (K), max.	550.28

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C527355&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C527355&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpcl:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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