

5-Fluoro-2-methylbenzoic acid

Other names:	Benzoic acid, 5-fluoro-2-methyl-
Inchi:	InChI=1S/C8H7FO2/c1-5-2-3-6(9)4-7(5)8(10)11/h2-4H,1H3,(H,10,11)
InchiKey:	JVBLXLBINTYFPR-UHFFFAOYSA-N
Formula:	C8H7FO2
SMILES:	<chem>Cc1ccc(F)cc1C(=O)O</chem>
Mol. weight [g/mol]:	154.14
CAS:	33184-16-6

Physical Properties

Property code	Value	Unit	Source
gf	-350.92	kJ/mol	Joback Method
hf	-455.78	kJ/mol	Joback Method
hfus	18.51	kJ/mol	Joback Method
hvap	59.61	kJ/mol	Joback Method
log10ws	-2.34		Crippen Method
logp	1.832		Crippen Method
mcvol	109.030	ml/mol	McGowan Method
pc	4031.24	kPa	Joback Method
tb	564.40	K	Joback Method
tc	763.02	K	Joback Method
tf	342.72	K	Joback Method
vc	0.418	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	243.15	J/molxK	564.40	Joback Method
cpg	251.74	J/molxK	597.50	Joback Method
cpg	259.85	J/molxK	630.61	Joback Method
cpg	267.50	J/molxK	663.71	Joback Method
cpg	274.70	J/molxK	696.82	Joback Method
cpg	281.46	J/molxK	729.92	Joback Method
cpg	287.81	J/molxK	763.02	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C33184166&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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