

1-Naphthoic acid, 2,2,3,3,4,4,5,5-octafluoropentyl ester

Inchi: InChI=1S/C16H10F8O2/c17-13(18)15(21,22)16(23,24)14(19,20)8-26-12(25)11-7-3-5-9-4
InchiKey: LFTALLNVWKRZAG-UHFFFAOYSA-N
Formula: C16H10F8O2
SMILES: O=C(OCC(F)(F)C(F)(F)C(F)(F)C(F)F)c1cccc2ccccc12
Mol. weight [g/mol]: 386.24

Physical Properties

Property code	Value	Unit	Source
gf	-1493.05	kJ/mol	Joback Method
hf	-1802.65	kJ/mol	Joback Method
hfus	29.53	kJ/mol	Joback Method
hvap	54.13	kJ/mol	Joback Method
log10ws	-6.45		Crippen Method
logp	5.168		Crippen Method
mcvol	214.680	ml/mol	McGowan Method
pc	1676.91	kPa	Joback Method
rinqol	1837.00		NIST Webbook
tb	676.44	K	Joback Method
tc	863.64	K	Joback Method
tf	410.86	K	Joback Method
vc	0.875	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	609.81	J/molxK	676.44	Joback Method
cpg	622.09	J/molxK	707.64	Joback Method
cpg	633.41	J/molxK	738.84	Joback Method
cpg	643.84	J/molxK	770.04	Joback Method
cpg	653.47	J/molxK	801.24	Joback Method
cpg	662.38	J/molxK	832.44	Joback Method
cpg	670.66	J/molxK	863.64	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U355684&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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