

Propionamide, 3-benzoyl-

Inchi:	InChI=1S/C10H11NO2/c11-10(13)7-6-9(12)8-4-2-1-3-5-8/h1-5H,6-7H2,(H2,11,13)
InchiKey:	PNPJKWAGJXUHIQ-UHFFFAOYSA-N
Formula:	C10H11NO2
SMILES:	NC(=O)CCC(=O)c1ccccc1
Mol. weight [g/mol]:	177.20
CAS:	23132-29-8

Physical Properties

Property code	Value	Unit	Source
gf	-45.66	kJ/mol	Joback Method
hf	-204.57	kJ/mol	Joback Method
hfus	24.09	kJ/mol	Joback Method
hvap	64.26	kJ/mol	Joback Method
log10ws	-2.18		Crippen Method
logp	1.135		Crippen Method
mcvol	141.120	ml/mol	McGowan Method
pc	3611.55	kPa	Joback Method
tb	635.15	K	Joback Method
tc	866.24	K	Joback Method
tf	412.00	K	Joback Method
vc	0.528	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	348.02	J/molxK	635.15	Joback Method
cpg	360.09	J/molxK	673.66	Joback Method
cpg	371.27	J/molxK	712.18	Joback Method
cpg	381.59	J/molxK	750.69	Joback Method
cpg	391.10	J/molxK	789.21	Joback Method
cpg	399.85	J/molxK	827.72	Joback Method
cpg	407.87	J/molxK	866.24	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C23132298&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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