

# 2-Butanone, oxime

<b>Other names:</b>	Butanone oxime Ethyl methyl ketone oxime Ethyl methyl ketoxime Methyl ethyl ketone oxime Methyl ethyl ketoxime MEK-Oxime USAF EK-906 Ethyl-methylketonoxim Skino No. 2 Troykyd anti-skin B USAF AM-3 2-Butoxime Aron M 1 SKINO 2 Pentan-2-one,oxime Butan-2-one, oxime NSC 442
<b>Inchi:</b>	InChI=1S/C4H9NO/c1-3-4(2)5-6/h6H,3H2,1-2H3
<b>InchiKey:</b>	WHIVNJATOVVLWBW-UHFFFAOYSA-N
<b>Formula:</b>	C4H9NO
<b>SMILES:</b>	CCC(C)=NO
<b>Mol. weight [g/mol]:</b>	87.12
<b>CAS:</b>	96-29-7

## Physical Properties

Property code	Value	Unit	Source
chs	-2727.00	kJ/mol	NIST Webbook
hf	-205.69	kJ/mol	Joback Method
hfs	-151.00	kJ/mol	NIST Webbook
hvap	57.70	kJ/mol	NIST Webbook
hvap	59.10 ± 0.20	kJ/mol	NIST Webbook
log10ws	-0.31		Crippen Method
logp	1.246		Crippen Method
mcpol	78.770	ml/mol	McGowan Method
pc	3857.88	kPa	Joback Method
rinpol	793.00		NIST Webbook
tb	425.70	K	NIST Webbook

tc	645.80	K	Joback Method
tf	243.65 ± 0.50	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hvapt	58.60 ± 0.20	kJ/mol	306.00	NIST Webbook
hvapt	53.70	kJ/mol	366.50	NIST Webbook
hvapt	55.50	kJ/mol	330.50	NIST Webbook
hvapt	57.20	kJ/mol	323.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	332.70	K	2.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C96297&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C96297&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

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