

# Ethyl 2-phenoxybenzoate

<b>Inchi:</b>	InChI=1S/C15H14O3/c1-2-17-15(16)13-10-6-7-11-14(13)18-12-8-4-3-5-9-12/h3-11H,2H2
<b>InchiKey:</b>	JGIQCXPHDHHHMN-UHFFFAOYSA-N
<b>Formula:</b>	C15H14O3
<b>SMILES:</b>	CCOC(=O)c1ccccc1Oc1ccccc1
<b>Mol. weight [g/mol]:</b>	242.27

## Physical Properties

Property code	Value	Unit	Source
gf	-48.31	kJ/mol	Joback Method
hf	-268.36	kJ/mol	Joback Method
hfus	26.27	kJ/mol	Joback Method
hvap	65.76	kJ/mol	Joback Method
log10ws	-3.77		Crippen Method
logp	3.656		Crippen Method
mcvol	188.000	ml/mol	McGowan Method
pc	2545.61	kPa	Joback Method
rinpol	1758.00		NIST Webbook
tb	699.65	K	Joback Method
tc	934.64	K	Joback Method
tf	418.56	K	Joback Method
vc	0.702	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	491.63	J/molxK	699.65	Joback Method
cpg	555.89	J/molxK	895.47	Joback Method
cpg	545.27	J/molxK	856.31	Joback Method
cpg	533.56	J/molxK	817.14	Joback Method
cpg	520.73	J/molxK	777.98	Joback Method
cpg	506.76	J/molxK	738.81	Joback Method
cpg	565.45	J/molxK	934.64	Joback Method
dvisc	0.0001064	Paxs	699.65	Joback Method
dvisc	0.0001337	Paxs	652.80	Joback Method

dvisc	0.0001741	Paxs	605.95	Joback Method
dvisc	0.0002370	Paxs	559.11	Joback Method
dvisc	0.0003412	Paxs	512.26	Joback Method
dvisc	0.0005287	Paxs	465.41	Joback Method
dvisc	0.0009037	Paxs	418.56	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R543023&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R543023&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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