

# DL-Glutamic acid

|                             |   |
|-----------------------------|---|
| <b>Other names:</b>         | Glutamic acid, DL-<br>(. +/-)-Glutamic acid<br>Glutamic acid<br>NSC 9967<br>Glutamic acid DL-form<br>2-Aminopropane-1,3-dicarboxylic acid<br>DL-glutamic acid monohydrate |
| <b>Inchi:</b>               | InChI=1S/C5H9NO4/c6-3(5(9)10)1-2-4(7)8/h3H,1-2,6H2,(H,7,8)(H,9,10)  |
| <b>InchiKey:</b>            | WHUUTDBJXRKMK-UHFFFAOYSA-N  |
| <b>Formula:</b>             | C5H9NO4   |
| <b>SMILES:</b>              | NC(CCC(=O)O)C(=O)O  |
| <b>Mol. weight [g/mol]:</b> | 147.13  |
| <b>CAS:</b>                 | 617-65-2  |

## Physical Properties

| Property code | Value    | Unit    | Source         |
|---------------|----------|---------|----------------|
| chs           | -2277.00 | kJ/mol  | NIST Webbook   |
| gf            | -476.25  | kJ/mol  | Joback Method  |
| hf            | -647.64  | kJ/mol  | Joback Method  |
| hfus          | 21.75    | kJ/mol  | Joback Method  |
| hvap          | 83.83    | kJ/mol  | Joback Method  |
| log10ws       | 0.34     |         | Crippen Method |
| logp          | -0.737   |         | Crippen Method |
| mcvol         | 106.170  | ml/mol  | McGowan Method |
| pc            | 5713.21  | kPa     | Joback Method  |
| tb            | 677.99   | K       | Joback Method  |
| tc            | 864.26   | K       | Joback Method  |
| tf            | 435.87   | K       | Joback Method  |
| vc            | 0.389    | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 279.71 | J/molxK | 677.99          | Joback Method |

|     |        |         |        |               |
|-----|--------|---------|--------|---------------|
| cpg | 286.05 | J/mol×K | 709.03 | Joback Method |
| cpg | 292.03 | J/mol×K | 740.08 | Joback Method |
| cpg | 297.65 | J/mol×K | 771.12 | Joback Method |
| cpg | 302.93 | J/mol×K | 802.17 | Joback Method |
| cpg | 307.87 | J/mol×K | 833.21 | Joback Method |
| cpg | 312.49 | J/mol×K | 864.26 | Joback Method |

## Sources

|                        |   |
|------------------------|---|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C617652&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C617652&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |

## Legend

|                 |   |
|-----------------|---|
| <b>chs:</b>     | Standard solid enthalpy of combustion           |
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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