

# Ethanone, 1-(2,4-dimethylphenyl)-

<b>Other names:</b>	Acetophenone, 2',4'-dimethyl- 2,4-Dimethylacetophenone 2',4'-Dimethylacetophenone 4-Acetyl-m-xylene (2,4-Dimethylphenyl)ethanone 2,4-(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub> -COCH <sub>3</sub> 1-(2,4-dimethylphenyl)ethanone
<b>Inchi:</b>	InChI=1S/C10H12O/c1-7-4-5-10(9(3)11)8(2)6-7/h4-6H,1-3H3
<b>InchiKey:</b>	HSDSKVWQTONQBJ-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>10</sub> H <sub>12</sub> O
<b>SMILES:</b>	CC(=O)c1ccc(C)cc1C
<b>Mol. weight [g/mol]:</b>	148.20
<b>CAS:</b>	89-74-7

## Physical Properties

Property code	Value	Unit	Source
affp	882.60	kJ/mol	NIST Webbook
basg	850.80	kJ/mol	NIST Webbook
gf	-2.45	kJ/mol	Joback Method
hf	-148.72	kJ/mol	Joback Method
hfus	16.52	kJ/mol	Joback Method
hvap	48.20	kJ/mol	Joback Method
log10ws	-3.08		Crippen Method
logp	2.506		Crippen Method
mcvol	129.570	ml/mol	McGowan Method
pc	3038.96	kPa	Joback Method
rinpol	1221.40		NIST Webbook
rinpol	1230.00		NIST Webbook
rinpol	1233.00		NIST Webbook
rinpol	1221.40		NIST Webbook
rinpol	1233.00		NIST Webbook
ripol	1809.00		NIST Webbook
ripol	1822.00		NIST Webbook
ripol	1876.00		NIST Webbook
ripol	1861.00		NIST Webbook
tb	501.20	K	NIST Webbook
tc	735.99	K	Joback Method

tf	296.10 ± 0.70	K	NIST Webbook
vc	0.493	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	276.59	J/mol×K	518.71	Joback Method
cpg	289.78	J/mol×K	554.92	Joback Method
cpg	302.26	J/mol×K	591.14	Joback Method
cpg	314.04	J/mol×K	627.35	Joback Method
cpg	325.13	J/mol×K	663.56	Joback Method
cpg	335.57	J/mol×K	699.77	Joback Method
cpg	345.38	J/mol×K	735.99	Joback Method
dvisc	0.0017002	Paxs	303.85	Joback Method
dvisc	0.0010355	Paxs	339.66	Joback Method
dvisc	0.0006932	Paxs	375.47	Joback Method
dvisc	0.0004977	Paxs	411.28	Joback Method
dvisc	0.0003768	Paxs	447.09	Joback Method
dvisc	0.0002973	Paxs	482.90	Joback Method
dvisc	0.0002423	Paxs	518.71	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	390.70	K	2.40	NIST Webbook

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C89747&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

# Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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