

# 2,4-Dichloro-1-methoxynaphthalene

<b>Inchi:</b>	InChI=1S/C11H8Cl2O/c1-14-11-8-5-3-2-4-7(8)9(12)6-10(11)13/h2-6H,1H3
<b>InchiKey:</b>	SMNLNHLWUBUPAT-UHFFFAOYSA-N
<b>Formula:</b>	C11H8Cl2O
<b>SMILES:</b>	COc1c(Cl)cc(Cl)c2ccccc12
<b>Mol. weight [g/mol]:</b>	227.09

## Physical Properties

Property code	Value	Unit	Source
gf	103.05	kJ/mol	Joback Method
hf	-40.88	kJ/mol	Joback Method
hfus	23.72	kJ/mol	Joback Method
hvap	57.16	kJ/mol	Joback Method
log10ws	-4.77		Crippen Method
logp	4.155		Crippen Method
mcvol	152.980	ml/mol	McGowan Method
pc	2973.04	kPa	Joback Method
rinpol	1670.00		NIST Webbook
tb	608.96	K	Joback Method
tc	850.96	K	Joback Method
tf	392.48	K	Joback Method
vc	0.582	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	322.39	J/molxK	608.96	Joback Method
cpg	333.75	J/molxK	649.29	Joback Method
cpg	344.30	J/molxK	689.63	Joback Method
cpg	354.09	J/molxK	729.96	Joback Method
cpg	363.17	J/molxK	770.29	Joback Method
cpg	371.58	J/molxK	810.62	Joback Method
cpg	379.36	J/molxK	850.96	Joback Method
dvisc	0.0009974	Paxs	392.48	Joback Method
dvisc	0.0007397	Paxs	428.56	Joback Method

dvisc	0.0005747	Paxs	464.64	Joback Method
dvisc	0.0004631	Paxs	500.72	Joback Method
dvisc	0.0003841	Paxs	536.80	Joback Method
dvisc	0.0003262	Paxs	572.88	Joback Method
dvisc	0.0002824	Paxs	608.96	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R540282&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R540282&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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