

# Ethanol, 2-(hexylamino)-

<b>Other names:</b>	2-hexylaminoethanol
<b>Inchi:</b>	InChI=1S/C8H19NO/c1-2-3-4-5-6-9-7-8-10/h9-10H,2-8H2,1H3
<b>InchiKey:</b>	MCIKGVLBLIZYRY-UHFFFAOYSA-N
<b>Formula:</b>	C8H19NO
<b>SMILES:</b>	CCCCCNCCO
<b>Mol. weight [g/mol]:</b>	145.24
<b>CAS:</b>	54596-69-9

## Physical Properties

Property code	Value	Unit	Source
gf	-30.95	kJ/mol	Joback Method
hf	-307.21	kJ/mol	Joback Method
hfus	25.66	kJ/mol	Joback Method
hvap	56.52	kJ/mol	Joback Method
log10ws	-1.62		Crippen Method
logp	1.149		Crippen Method
mcvol	139.430	ml/mol	McGowan Method
pc	2826.33	kPa	Joback Method
rinpol	1197.00		NIST Webbook
rinpol	1197.00		NIST Webbook
tb	524.79	K	Joback Method
tc	688.02	K	Joback Method
tf	293.40	K	Joback Method
vc	0.537	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	338.39	J/molxK	524.79	Joback Method
cpg	350.27	J/molxK	552.00	Joback Method
cpg	361.68	J/molxK	579.20	Joback Method
cpg	372.63	J/molxK	606.41	Joback Method
cpg	383.14	J/molxK	633.61	Joback Method
cpg	393.21	J/molxK	660.82	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	394.50 ± 0.50	K	1.90	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C54596699&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C54596699&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/66-040-4/Ethanol-2-hexylamino.pdf>

Generated by Cheméo on 2024-04-27 21:23:28.847808595 +0000 UTC m=+16542257.768385911.  
Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.