

# 2,5-Dihydro-3,4-dimethylthiophen-2-one

<b>Other names:</b>	2,5-Dihydrothiophen-2-one, 3,4-dimethyl 2(5H)-Thiophenone,3,4-dimethyl-
<b>Inchi:</b>	InChI=1S/C6H8OS/c1-4-3-8-6(7)5(4)2/h3H2,1-2H3
<b>InchiKey:</b>	IXXRQTNARJNVHF-UHFFFAOYSA-N
<b>Formula:</b>	C6H8OS
<b>SMILES:</b>	CC1=C(C)C(=O)SC1
<b>Mol. weight [g/mol]:</b>	128.19
<b>CAS:</b>	33922-75-7

## Physical Properties

Property code	Value	Unit	Source
gf	-28.13	kJ/mol	Joback Method
hf	-143.95	kJ/mol	Joback Method
hfus	7.77	kJ/mol	Joback Method
hvap	41.19	kJ/mol	Joback Method
ie	9.44 ± 0.05	eV	NIST Webbook
log10ws	-1.74		Crippen Method
logp	1.596		Crippen Method
mcvol	98.160	ml/mol	McGowan Method
pc	4183.90	kPa	Joback Method
rinpol	1193.00		NIST Webbook
rinpol	1217.00		NIST Webbook
rinpol	1200.00		NIST Webbook
tb	481.40	K	Joback Method
tc	720.36	K	Joback Method
tf	349.99	K	Joback Method
vc	0.352	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	191.02	J/mol×K	481.40	Joback Method
cpg	201.89	J/mol×K	521.23	Joback Method
cpg	212.30	J/mol×K	561.05	Joback Method

cpg	222.23	J/mol×K	600.88	Joback Method
cpg	231.67	J/mol×K	640.71	Joback Method
cpg	240.62	J/mol×K	680.54	Joback Method
cpg	249.07	J/mol×K	720.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C33922757&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C33922757&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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