

2,2,2'-Trichloro-n-butyl propionanilide

Inchi:	InChI=1S/C13H16Cl3NO/c1-3-4-9-17(12(18)13(2,15)16)11-8-6-5-7-10(11)14/h5-8H,3-4,9
InchiKey:	FRYHGFZDEVTUQV-UHFFFAOYSA-N
Formula:	C13H16Cl3NO
SMILES:	CCCCN(C(=O)C(C)(Cl)Cl)c1ccccc1Cl
Mol. weight [g/mol]:	308.63
CAS:	116402-44-9

Physical Properties

Property code	Value	Unit	Source
gf	110.27	kJ/mol	Joback Method
hf	-187.61	kJ/mol	Joback Method
hfus	32.87	kJ/mol	Joback Method
hvap	68.12	kJ/mol	Joback Method
log10ws	-4.85		Crippen Method
logp	4.667		Crippen Method
mcvol	218.540	ml/mol	McGowan Method
pc	2102.27	kPa	Joback Method
tb	703.87	K	Joback Method
tc	928.95	K	Joback Method
tf	449.79	K	Joback Method
vc	0.816	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	539.16	J/molxK	703.87	Joback Method
cpg	552.64	J/molxK	741.38	Joback Method
cpg	565.08	J/molxK	778.90	Joback Method
cpg	576.56	J/molxK	816.41	Joback Method
cpg	587.16	J/molxK	853.92	Joback Method
cpg	596.99	J/molxK	891.43	Joback Method
cpg	606.11	J/molxK	928.95	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C116402449&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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