

# Diethylmalonic acid, butyl 2-isopropoxyphenyl ester

**Inchi:** InChI=1S/C20H30O5/c1-6-9-14-23-18(21)20(7-2,8-3)19(22)25-17-13-11-10-12-16(17)24  
**InchiKey:** GAUYBNIAYCWWOG-UHFFFAOYSA-N  
**Formula:** C20H30O5  
**SMILES:** CCCOC(=O)C(CC)(CC)C(=O)Oc1ccccc1OC(C)C  
**Mol. weight [g/mol]:** 350.45

## Physical Properties

Property code	Value	Unit	Source
gf	-352.14	kJ/mol	Joback Method
hf	-866.92	kJ/mol	Joback Method
hfus	37.03	kJ/mol	Joback Method
hvap	82.09	kJ/mol	Joback Method
log10ws	-5.24		Crippen Method
logp	4.529		Crippen Method
mvol	289.650	ml/mol	McGowan Method
pc	1351.64	kPa	Joback Method
rinpol	2155.00		NIST Webbook
rinpol	2155.00		NIST Webbook
tb	859.99	K	Joback Method
tc	1067.72	K	Joback Method
tf	508.07	K	Joback Method
vc	1.097	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	908.47	J/molxK	859.99	Joback Method
cpg	924.35	J/molxK	894.61	Joback Method
cpg	938.98	J/molxK	929.23	Joback Method
cpg	952.39	J/molxK	963.86	Joback Method
cpg	964.61	J/molxK	998.48	Joback Method
cpg	975.68	J/molxK	1033.10	Joback Method
cpg	985.60	J/molxK	1067.72	Joback Method
dvisc	0.0004263	Paxs	508.07	Joback Method

dvisc	0.0002171	Paxs	566.72	Joback Method
dvisc	0.0001255	Paxs	625.38	Joback Method
dvisc	0.0000797	Paxs	684.03	Joback Method
dvisc	0.0000543	Paxs	742.68	Joback Method
dvisc	0.0000392	Paxs	801.34	Joback Method
dvisc	0.0000296	Paxs	859.99	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369584&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369584&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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