

# 2-chlorohexyl dichloroacetate

<b>Other names:</b>	1-Hexanol, 2-chloro, dichloroacetate
<b>Inchi:</b>	InChI=1S/C8H13Cl3O2/c1-2-3-4-6(9)5-13-8(12)7(10)11/h6-7H,2-5H2,1H3
<b>InchiKey:</b>	PFUJZHGRURGIP-UHFFFAOYSA-N
<b>Formula:</b>	C8H13Cl3O2
<b>SMILES:</b>	CCCCC(Cl)COC(=O)C(Cl)Cl
<b>Mol. weight [g/mol]:</b>	247.55

## Physical Properties

Property code	Value	Unit	Source
gf	-258.11	kJ/mol	Joback Method
hf	-511.03	kJ/mol	Joback Method
hfus	24.81	kJ/mol	Joback Method
hvap	54.94	kJ/mol	Joback Method
log10ws	-3.21		Crippen Method
logp	3.131		Crippen Method
mcvol	167.740	ml/mol	McGowan Method
pc	2414.74	kPa	Joback Method
rinpol	1396.00		NIST Webbook
rinpol	1418.00		NIST Webbook
rinpol	1396.00		NIST Webbook
rinpol	1412.00		NIST Webbook
rinpol	1409.00		NIST Webbook
rinpol	1425.00		NIST Webbook
ripol	2044.00		NIST Webbook
ripol	2047.00		NIST Webbook
ripol	2037.00		NIST Webbook
tb	570.14	K	Joback Method
tc	768.49	K	Joback Method
tf	311.84	K	Joback Method
vc	0.642	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	360.64	J/molxK	570.14	Joback Method
cpg	411.57	J/molxK	735.43	Joback Method
cpg	402.51	J/molxK	702.37	Joback Method
cpg	392.89	J/molxK	669.32	Joback Method
cpg	382.72	J/molxK	636.26	Joback Method
cpg	371.97	J/molxK	603.20	Joback Method
cpg	420.09	J/molxK	768.49	Joback Method
dvisc	0.0002110	Paxs	570.14	Joback Method
dvisc	0.0002819	Paxs	527.09	Joback Method
dvisc	0.0003965	Paxs	484.04	Joback Method
dvisc	0.0005961	Paxs	440.99	Joback Method
dvisc	0.0009788	Paxs	397.94	Joback Method
dvisc	0.0018126	Paxs	354.89	Joback Method
dvisc	0.0039796	Paxs	311.84	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R111923&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R111923&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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