

# Thioacetic acid

<b>Other names:</b>	Methanecarbothiolic acid Thioacetic S-acid Ethanethiolic acid Acetyl mercaptan Ethanethiolic acid Thiacetic acid Thiolacetic acid Thionoacetic acid CH <sub>3</sub> COSH Kyselina thiooctova UN 2436 USAF EK-P-737 Acetic acid, thio-
<b>Inchi:</b>	InChI=1S/C2H4OS/c1-2(3)4/h1H3,(H,3,4)
<b>InchiKey:</b>	DUYAAUVXQSMXQP-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>2</sub> H <sub>4</sub> OS
<b>SMILES:</b>	CC(=O)S
<b>Mol. weight [g/mol]:</b>	76.12
<b>CAS:</b>	507-09-5

## Physical Properties

Property code	Value	Unit	Source
chl	-1741.50	kJ/mol	NIST Webbook
gf	-133.57	kJ/mol	Joback Method
hf	-158.71	kJ/mol	Joback Method
hfl	-216.50 ± 0.59	kJ/mol	NIST Webbook
hfl	-219.30 ± 1.50	kJ/mol	NIST Webbook
hfus	6.58	kJ/mol	Joback Method
hvap	33.53	kJ/mol	Joback Method
ie	10.06	eV	NIST Webbook
log10ws	-0.51		Crippen Method
logp	0.463		Crippen Method
mcvol	56.960	ml/mol	McGowan Method
pc	6075.01	kPa	Joback Method
ripol	1124.00		NIST Webbook
ripol	1124.00		NIST Webbook
tb	361.89	K	Joback Method

tc	570.61	K	Joback Method
tf	198.69	K	Joback Method
vc	0.207	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	102.74	J/mol×K	535.82	Joback Method
cpg	82.03	J/mol×K	361.89	Joback Method
cpg	86.54	J/mol×K	396.68	Joback Method
cpg	90.87	J/mol×K	431.46	Joback Method
cpg	95.01	J/mol×K	466.25	Joback Method
cpg	98.96	J/mol×K	501.03	Joback Method
cpg	106.34	J/mol×K	570.61	Joback Method
hvapt	35.20	kJ/mol	333.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C507095&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C507095&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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