

# 2-Pentyl-2H-thiapyrane

<b>Other names:</b>	2-pentyl-(2H)-thiapyran 2-pentylthiapyran
<b>Inchi:</b>	InChI=1S/C10H16S/c1-2-3-4-7-10-8-5-6-9-11-10/h5-6,8-10H,2-4,7H2,1H3
<b>InchiKey:</b>	WIWJPSVDDXGSQF-UHFFFAOYSA-N
<b>Formula:</b>	C10H16S
<b>SMILES:</b>	CCCCC1C=CC=CS1
<b>Mol. weight [g/mol]:</b>	168.30

## Physical Properties

Property code	Value	Unit	Source
gf	157.55	kJ/mol	Joback Method
hf	-34.59	kJ/mol	Joback Method
hfus	19.59	kJ/mol	Joback Method
hvap	44.68	kJ/mol	Joback Method
log10ws	-4.10		Crippen Method
logp	3.752		Crippen Method
mcvol	148.650	ml/mol	McGowan Method
pc	2749.78	kPa	Joback Method
rinpol	1344.00		NIST Webbook
rinpol	1317.00		NIST Webbook
rinpol	1328.00		NIST Webbook
rinpol	1329.00		NIST Webbook
rinpol	1337.00		NIST Webbook
ripol	1700.00		NIST Webbook
tb	493.90	K	Joback Method
tc	708.45	K	Joback Method
tf	294.81	K	Joback Method
vc	0.546	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	319.98	J/mol×K	493.90	Joback Method
cpg	336.99	J/mol×K	529.66	Joback Method

cpg	353.02	J/mol×K	565.42	Joback Method
cpg	368.12	J/mol×K	601.17	Joback Method
cpg	382.32	J/mol×K	636.93	Joback Method
cpg	395.66	J/mol×K	672.69	Joback Method
cpg	408.17	J/mol×K	708.45	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R180479&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R180479&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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