

# Acetic acid, (4-chlorophenoxy)-

|                             |  |
|-----------------------------|--|
| <b>Other names:</b>         | (4-Chlorophenoxy)acetic acid<br>(p-Chlorophenoxy)acetic acid<br>2-(4-chlorophenoxy)acetic acid<br>4-CP<br>4-CPA<br>Acetic acid, (p-chlorophenoxy)-<br>Acetic acid, 2-(4-chlorophenoxy)-<br>BI 12<br>CPA<br>Kyselina 4-chlorfenoxyoctova<br>Marks 4-cpa<br>NSC 8769<br>PCPA<br>Sure-Set<br>Tomato Fix<br>Tomato Fix concentrate<br>Tomato hold<br>Tomatotone<br>para-Chlorophenoxyacetic acid |
| <b>Inchi:</b>               | InChI=1S/C8H7ClO3/c9-6-1-3-7(4-2-6)12-5-8(10)11/h1-4H,5H2,(H,10,11)  |
| <b>InchiKey:</b>            | SODPIMGUZLOIPE-UHFFFAOYSA-N  |
| <b>Formula:</b>             | C8H7ClO3   |
| <b>SMILES:</b>              | O=C(O)COc1ccc(Cl)cc1   |
| <b>Mol. weight [g/mol]:</b> | 186.59   |
| <b>CAS:</b>                 | 122-88-3   |

## Physical Properties

| Property code | Value   | Unit   | Source                               |
|---------------|---------|--------|--------------------------------------|
| gf            | -263.41 | kJ/mol | Joback Method                        |
| hf            | -396.16 | kJ/mol | Joback Method                        |
| hfus          | 21.20   | kJ/mol | Joback Method                        |
| hvap          | 66.56   | kJ/mol | Joback Method                        |
| log10ws       | -2.29   |        | Aqueous Solubility Prediction Method |
| logp          | 1.803   |        | Crippen Method                       |
| mvol          | 125.370 | ml/mol | McGowan Method                       |
| pc            | 4062.13 | kPa    | Joback Method                        |

|    |               |                      |                                      |
|----|---------------|----------------------|--------------------------------------|
| tb | 620.00        | K                    | Joback Method                        |
| tc | 829.16        | K                    | Joback Method                        |
| tf | 434.65        | K                    | Aqueous Solubility Prediction Method |
| tf | 431.48 ± 0.20 | K                    | NIST Webbook                         |
| vc | 0.468         | m <sup>3</sup> /kmol | Joback Method                        |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 316.43    | J/mol×K | 794.30          | Joback Method |
| cpg           | 309.96    | J/mol×K | 759.44          | Joback Method |
| cpg           | 302.98    | J/mol×K | 724.58          | Joback Method |
| cpg           | 295.49    | J/mol×K | 689.72          | Joback Method |
| cpg           | 287.49    | J/mol×K | 654.86          | Joback Method |
| cpg           | 278.94    | J/mol×K | 620.00          | Joback Method |
| cpg           | 322.41    | J/mol×K | 829.16          | Joback Method |
| dvisc         | 0.0022303 | Paxs    | 381.76          | Joback Method |
| dvisc         | 0.0000801 | Paxs    | 620.00          | Joback Method |
| dvisc         | 0.0001153 | Paxs    | 580.29          | Joback Method |
| dvisc         | 0.0001753 | Paxs    | 540.59          | Joback Method |
| dvisc         | 0.0002846 | Paxs    | 500.88          | Joback Method |
| dvisc         | 0.0005023 | Paxs    | 461.17          | Joback Method |
| dvisc         | 0.0009866 | Paxs    | 421.47          | Joback Method |
| hfust         | 36.27     | kJ/mol  | 429.60          | NIST Webbook  |
| hfust         | 36.27     | kJ/mol  | 429.60          | NIST Webbook  |

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C122883&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

# Legend

|                                       |   |
|---------------------------------------|---|
| <b>cp<sub>g</sub>:</b>                | Ideal gas heat capacity                         |
| <b>d<sub>visc</sub>:</b>              | Dynamic viscosity                               |
| <b>g<sub>f</sub>:</b>                 | Standard Gibbs free energy of formation         |
| <b>h<sub>f</sub>:</b>                 | Enthalpy of formation at standard conditions    |
| <b>h<sub>fus</sub>:</b>               | Enthalpy of fusion at standard conditions       |
| <b>h<sub>fust</sub>:</b>              | Enthalpy of fusion at a given temperature       |
| <b>h<sub>vap</sub>:</b>               | Enthalpy of vaporization at standard conditions |
| <b>log<sub>10</sub>w<sub>s</sub>:</b> | Log <sub>10</sub> of Water solubility in mol/l  |
| <b>log<sub>p</sub>:</b>               | Octanol/Water partition coefficient             |
| <b>mc<sub>vol</sub>:</b>              | McGowan's characteristic volume                 |
| <b>p<sub>c</sub>:</b>                 | Critical Pressure                               |
| <b>t<sub>b</sub>:</b>                 | Normal Boiling Point Temperature                |
| <b>t<sub>c</sub>:</b>                 | Critical Temperature                            |
| <b>t<sub>f</sub>:</b>                 | Normal melting (fusion) point                   |
| <b>v<sub>c</sub>:</b>                 | Critical Volume                                 |

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