

# 1,3-Butanedithiol

<b>Other names:</b>	1,3-dimercaptobutane butane-1,3-dithiol
<b>Inchi:</b>	InChI=1S/C4H10S2/c1-4(6)2-3-5/h4-6H,2-3H2,1H3
<b>InchiKey:</b>	XMEPRJBZFCWFKN-UHFFFAOYSA-N
<b>Formula:</b>	C4H10S2
<b>SMILES:</b>	CC(S)CCS
<b>Mol. weight [g/mol]:</b>	122.25
<b>CAS:</b>	24330-52-7

## Physical Properties

Property code	Value	Unit	Source
gf	39.14	kJ/mol	Joback Method
hf	-54.21	kJ/mol	Joback Method
hfus	10.68	kJ/mol	Joback Method
hvap	37.58	kJ/mol	Joback Method
log10ws	-1.75		Crippen Method
logp	1.625		Crippen Method
mcvol	99.920	ml/mol	McGowan Method
pc	4652.99	kPa	Joback Method
rinp	995.00		NIST Webbook
rinp	995.00		NIST Webbook
rip	1456.00		NIST Webbook
tb	416.20	K	Joback Method
tc	639.97	K	Joback Method
tf	192.76	K	Joback Method
vc	0.361	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	173.79	J/molxK	416.20	Joback Method
cpg	183.51	J/molxK	453.50	Joback Method
cpg	192.74	J/molxK	490.79	Joback Method
cpg	201.50	J/molxK	528.09	Joback Method

cpg	209.78	J/mol×K	565.38	Joback Method
cpg	217.62	J/mol×K	602.68	Joback Method
cpg	225.02	J/mol×K	639.97	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.35882e+01
Coeff. B	-3.60799e+03
Coeff. C	-6.86550e+01
Temperature range (K), min.	339.92
Temperature range (K), max.	504.58

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24330527&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24330527&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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