

# 2-Propenenitrile, 3-phenyl-, (E)-

<b>Other names:</b>	trans-Cinnamonnitrile (E)-Cinnamonnitrile (E)-3-Phenylacrylonitrile trans-«beta»-Phenylacrylonitrile trans-3-Phenylpropenenitrile Cinnamonnitrile, (E)- (E)-3-Phenylpropenenitrile (E)-3-Phenylprop-2-enenitrile 2-Propenenitrile, 3-phenyl-, (2E)- NSC 77496 trans-3-Phenyl-2-propenenitrile cinnamonnitrile
<b>Inchi:</b>	InChI=1S/C9H7N/c10-8-4-7-9-5-2-1-3-6-9/h1-7H/b7-4+
<b>InchiKey:</b>	ZWKNLRFUTWSOY-QPJJXVBHSA-N
<b>Formula:</b>	C9H7N
<b>SMILES:</b>	N#CC=Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	129.16
<b>CAS:</b>	1885-38-7

## Physical Properties

Property code	Value	Unit	Source
gf	350.71	kJ/mol	Joback Method
hf	289.54	kJ/mol	Joback Method
hfus	14.81	kJ/mol	Joback Method
hvap	48.34	kJ/mol	Joback Method
log10ws	-2.58		Crippen Method
logp	2.223		Crippen Method
mcvol	110.990	ml/mol	McGowan Method
pc	3364.54	kPa	Joback Method
rinpol	1301.10		NIST Webbook
rinpol	1301.10		NIST Webbook
tb	527.70	K	NIST Webbook
tc	776.63	K	Joback Method
tf	277.52	K	Joback Method
vc	0.438	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	225.13	J/molxK	538.24	Joback Method
cpg	235.97	J/molxK	577.97	Joback Method
cpg	245.97	J/molxK	617.70	Joback Method
cpg	255.18	J/molxK	657.44	Joback Method
cpg	263.67	J/molxK	697.17	Joback Method
cpg	271.49	J/molxK	736.90	Joback Method
cpg	278.72	J/molxK	776.63	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1885387&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1885387&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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