

# 3,5-di-tert-Butyl-4-hydroxyanisole

<b>Other names:</b>	3,5-di-t-Butyl-4-hydroxyanisole Phenol, 2,6-bis(1,1-dimethylethyl)-4-methoxy- Phenol, 2,6-di-tert-butyl-4-methoxy- Topanol 354 2,6-Bis(1,1-dimethylethyl)-4-methoxyphenol 2,6-Di-tert-butyl-4-methoxyphenol Phenol, 4-methoxy, 2,6-bis-(1,1-dimethylethyl)
<b>Inchi:</b>	InChI=1S/C15H24O2/c1-14(2,3)11-8-10(17-7)9-12(13(11)16)15(4,5)6/h8-9,16H,1-7H3
<b>InchiKey:</b>	SLUKQUGVTITNSY-UHFFFAOYSA-N
<b>Formula:</b>	C15H24O2
<b>SMILES:</b>	COc1cc(C(C)(C)C)c(O)c(C(C)(C)C)c1
<b>Mol. weight [g/mol]:</b>	236.35
<b>CAS:</b>	489-01-0

## Physical Properties

Property code	Value	Unit	Source
gf	-85.37	kJ/mol	Joback Method
hf	-466.37	kJ/mol	Joback Method
hfus	20.01	kJ/mol	Joback Method
hvap	65.42	kJ/mol	Joback Method
log10ws	-3.77		Crippen Method
logp	3.996		Crippen Method
mvol	210.190	ml/mol	McGowan Method
pc	2090.75	kPa	Joback Method
rinpol	1527.00		NIST Webbook
tb	675.82	K	Joback Method
tc	899.57	K	Joback Method
tf	449.06	K	Joback Method
vc	0.730	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	679.82	J/mol×K	899.57	Joback Method

cpg	589.88	J/mol×K	675.82	Joback Method
cpg	607.26	J/mol×K	713.11	Joback Method
cpg	623.52	J/mol×K	750.40	Joback Method
cpg	638.80	J/mol×K	787.69	Joback Method
cpg	653.20	J/mol×K	824.99	Joback Method
cpg	666.84	J/mol×K	862.28	Joback Method
dvisc	0.0000096	Paxs	675.82	Joback Method
dvisc	0.0003388	Paxs	449.06	Joback Method
dvisc	0.0001485	Paxs	486.85	Joback Method
dvisc	0.0000733	Paxs	524.65	Joback Method
dvisc	0.0000398	Paxs	562.44	Joback Method
dvisc	0.0000233	Paxs	600.23	Joback Method
dvisc	0.0000146	Paxs	638.03	Joback Method
hfust	26.90	kJ/mol	374.40	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C489010&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C489010&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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