

3,3-Diethyloctane

Inchi:	InChI=1S/C12H26/c1-5-9-10-11-12(6-2,7-3)8-4/h5-11H2,1-4H3
InchiKey:	DGJISSKLLWWXTG-UHFFFAOYSA-N
Formula:	C12H26
SMILES:	CCCCC(CC)(CC)CC
Mol. weight [g/mol]:	170.33
CAS:	17302-40-8

Physical Properties

Property code	Value	Unit	Source
gf	53.00	kJ/mol	Joback Method
hf	-299.76	kJ/mol	Joback Method
hfus	19.42	kJ/mol	Joback Method
hvap	41.01	kJ/mol	Joback Method
log10ws	-4.60		Crippen Method
logp	4.783		Crippen Method
mcvol	179.940	ml/mol	McGowan Method
pc	1809.23	kPa	Joback Method
rinpol	1167.00		NIST Webbook
rinpol	1167.00		NIST Webbook
tb	470.73	K	Joback Method
tc	640.99	K	Joback Method
tf	227.42	K	Joback Method
vc	0.697	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	408.96	J/mol×K	470.73	Joback Method
cpg	491.58	J/mol×K	612.61	Joback Method
cpg	476.54	J/mol×K	584.24	Joback Method
cpg	460.79	J/mol×K	555.86	Joback Method
cpg	444.30	J/mol×K	527.48	Joback Method
cpg	427.03	J/mol×K	499.11	Joback Method
cpg	505.93	J/mol×K	640.99	Joback Method

dvisc	0.0002128	Paxs	470.73	Joback Method
dvisc	0.0002979	Paxs	430.18	Joback Method
dvisc	0.0004474	Paxs	389.63	Joback Method
dvisc	0.0007385	Paxs	349.07	Joback Method
dvisc	0.0013906	Paxs	308.52	Joback Method
dvisc	0.0031716	Paxs	267.97	Joback Method
dvisc	0.0097058	Paxs	227.42	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.39478e+01
Coeff. B	-3.60895e+03
Coeff. C	-9.13150e+01
Temperature range (K), min.	355.51
Temperature range (K), max.	509.20

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R522848&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
m_{cvol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
r_{inpol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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