

Disulfide, methyl phenyl

Other names:	Methyl phenyl disulfide Methyldisulfanylbenezene methyl phenyl disulphide
Inchi:	InChI=1S/C7H8S2/c1-8-9-7-5-3-2-4-6-7/h2-6H,1H3
InchiKey:	LMSQHVVXHZCNJEP-UHFFFAOYSA-N
Formula:	C7H8S2
SMILES:	CSSc1ccccc1
Mol. weight [g/mol]:	156.27
CAS:	14173-25-2

Physical Properties

Property code	Value	Unit	Source
gf	186.71	kJ/mol	Joback Method
hf	132.46	kJ/mol	Joback Method
hfus	16.19	kJ/mol	Joback Method
hvap	47.09	kJ/mol	Joback Method
log10ws	-3.09		Crippen Method
logp	3.057		Crippen Method
mcvol	118.430	ml/mol	McGowan Method
pc	4299.92	kPa	Joback Method
rinpol	1303.00		NIST Webbook
tb	523.80	K	Joback Method
tc	787.37	K	Joback Method
tf	263.87	K	Joback Method
vc	0.427	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	226.59	J/mol×K	523.80	Joback Method
cpg	238.91	J/mol×K	567.73	Joback Method
cpg	250.33	J/mol×K	611.66	Joback Method
cpg	260.86	J/mol×K	655.59	Joback Method
cpg	270.53	J/mol×K	699.51	Joback Method

cpg	279.35	J/mol×K	743.44	Joback Method
cpg	287.35	J/mol×K	787.37	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C14173252&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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