

# Dimethylmalonic acid, monochloride, 3-phenylpropyl ester

Inchi:	InChI=1S/C14H17ClO3/c1-14(2,12(15)16)13(17)18-10-6-9-11-7-4-3-5-8-11/h3-5,7-8H,6,9
InchiKey:	BMNDSPKXQWDZTK-UHFFFAOYSA-N
Formula:	C14H17ClO3
SMILES:	CC(C)(C(=O)Cl)C(=O)OCCc1ccccc1
Mol. weight [g/mol]:	268.74

## Physical Properties

Property code	Value	Unit	Source
gf	-192.52	kJ/mol	Joback Method
hf	-477.63	kJ/mol	Joback Method
hfus	27.23	kJ/mol	Joback Method
hvap	68.02	kJ/mol	Joback Method
log10ws	-3.34		Crippen Method
logp	2.954		Crippen Method
mcvol	205.610	ml/mol	McGowan Method
pc	2183.60	kPa	Joback Method
rinpol	1798.00		NIST Webbook
tb	710.76	K	Joback Method
tc	931.73	K	Joback Method
tf	428.39	K	Joback Method
vc	0.779	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	543.60	J/molxK	710.76	Joback Method
cpg	557.76	J/molxK	747.59	Joback Method
cpg	570.86	J/molxK	784.42	Joback Method
cpg	582.96	J/molxK	821.24	Joback Method
cpg	594.12	J/molxK	858.07	Joback Method
cpg	604.40	J/molxK	894.90	Joback Method
cpg	613.84	J/molxK	931.73	Joback Method
dvisc	0.0014179	Paxs	428.39	Joback Method
dvisc	0.0007618	Paxs	475.45	Joback Method

dvisc	0.0004578	Paxs	522.51	Joback Method
dvisc	0.0002993	Paxs	569.58	Joback Method
dvisc	0.0002087	Paxs	616.64	Joback Method
dvisc	0.0001532	Paxs	663.70	Joback Method
dvisc	0.0001172	Paxs	710.76	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361848&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361848&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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