

L-Leucine, N-ethoxycarbonyl-N-methyl-, undecyl ester

Inchi: InChI=1S/C21H41NO4/c1-6-8-9-10-11-12-13-14-15-16-26-20(23)19(17-18(3)4)22(5)21(2)
InchiKey: IIEFYFJXUPPNIO-UHFFFAOYSA-N
Formula: C21H41NO4
SMILES: CCCCCCCCCCOC(=O)C(CC(C)C)N(C)C(=O)OCC
Mol. weight [g/mol]: 371.56

Physical Properties

Property code	Value	Unit	Source
hf	-1055.81	kJ/mol	Cheméo Relay (1.0)
hfus	51.69	kJ/mol	Joback Method
hvap	105.98	kJ/mol	Cheméo Relay (1.0)
log10ws	-5.22		Cheméo Relay (1.0)
logp	5.563		Crippen Method
mcvol	331.610	ml/mol	McGowan Method
rinpol	2324.00		NIST Webbook
tb	618.70	K	Cheméo Relay (1.0)
tf	267.80	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1074.95	J/mol×K	844.02	Joback Method
cpg	1093.87	J/mol×K	875.71	Joback Method
cpg	1111.61	J/mol×K	907.40	Joback Method
cpg	1128.18	J/mol×K	939.09	Joback Method
cpg	1143.62	J/mol×K	970.78	Joback Method
cpg	1157.96	J/mol×K	1002.48	Joback Method
cpg	1171.23	J/mol×K	1034.17	Joback Method

Sources

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=U321926&Units=SI>

Joback Method: https://en.wikipedia.org/wiki/Joback_method

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tf:	Normal melting (fusion) point

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