

(E)-4-Methylhept-3-ene

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|-----------------------------|---|
| Other names: | 3-Heptene, 4-methyl-, trans |
| Inchi: | InChI=1S/C8H16/c1-4-6-8(3)7-5-2/h6H,4-5,7H2,1-3H3/b8-6+ |
| InchiKey: | KKVVJQGDNYIIMN-SOFGYWHQSA-N |
| Formula: | C8H16 |
| SMILES: | CCC=C(C)CCC |
| Mol. weight [g/mol]: | 112.21 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | 88.15 | kJ/mol | Joback Method |
| hf | -101.02 | kJ/mol | Joback Method |
| hfus | 15.37 | kJ/mol | Joback Method |
| hvap | 33.44 | kJ/mol | Joback Method |
| log10ws | -3.02 | | Crippen Method |
| logp | 3.143 | | Crippen Method |
| mcvol | 119.280 | ml/mol | McGowan Method |
| pc | 2687.42 | kPa | Joback Method |
| rinpol | 791.30 | | NIST Webbook |
| rinpol | 791.00 | | NIST Webbook |
| rinpol | 791.00 | | NIST Webbook |
| rinpol | 779.00 | | NIST Webbook |
| rinpol | 778.00 | | NIST Webbook |
| rinpol | 791.00 | | NIST Webbook |
| rinpol | 791.30 | | NIST Webbook |
| tb | 386.48 | K | Joback Method |
| tc | 561.75 | K | Joback Method |
| tf | 160.88 | K | Joback Method |
| vc | 0.465 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 218.03 | J/mol×K | 386.48 | Joback Method |
| cpg | 231.15 | J/mol×K | 415.69 | Joback Method |

| | | | | |
|-----|--------|---------|--------|---------------|
| cpg | 243.70 | J/mol×K | 444.90 | Joback Method |
| cpg | 255.70 | J/mol×K | 474.11 | Joback Method |
| cpg | 267.18 | J/mol×K | 503.32 | Joback Method |
| cpg | 278.15 | J/mol×K | 532.54 | Joback Method |
| cpg | 288.64 | J/mol×K | 561.75 | Joback Method |

Sources

| | |
|------------------------|---|
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=R293040&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |

Legend

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|------------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| h vap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| m cvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| r in pol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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<https://www.chemeo.com/cid/66-600-2/E-4-Methylhept-3-ene.pdf>

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