

# Butanedioic acid, acetyl-, diethyl ester

<b>Other names:</b>	Diethyl acetylsuccinate Acetylsuccinic acid diethyl ester Diethyl acetosuccinate Diethyl 2-acetylsuccinate Ethyl acetylsuccinate Succinic acid, acetyl-, diethyl ester
<b>Inchi:</b>	InChI=1S/C10H16O5/c1-4-14-9(12)6-8(7(3)11)10(13)15-5-2/h8H,4-6H2,1-3H3
<b>InchiKey:</b>	DVSDDICSXBCMQL-UHFFFAOYSA-N
<b>Formula:</b>	C10H16O5
<b>SMILES:</b>	CCOC(=O)CC(C(C)=O)C(=O)OCC
<b>Mol. weight [g/mol]:</b>	216.23
<b>CAS:</b>	1115-30-6

## Physical Properties

Property code	Value	Unit	Source
gf	-565.88	kJ/mol	Joback Method
hf	-857.19	kJ/mol	Joback Method
hfus	25.31	kJ/mol	Joback Method
hvap	62.52	kJ/mol	Joback Method
log10ws	-0.77		Crippen Method
logp	0.708		Crippen Method
mcvol	168.210	ml/mol	McGowan Method
pc	2462.92	kPa	Joback Method
tb	528.20	K	NIST Webbook
tb	528.00 ± 1.00	K	NIST Webbook
tc	825.12	K	Joback Method
tf	381.71	K	Joback Method
vc	0.643	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	434.27	J/mol×K	634.21	Joback Method
cpg	446.78	J/mol×K	666.03	Joback Method

cpg	458.67	J/molxK	697.85	Joback Method
cpg	469.95	J/molxK	729.67	Joback Method
cpg	480.59	J/molxK	761.48	Joback Method
cpg	490.60	J/molxK	793.30	Joback Method
cpg	499.97	J/molxK	825.12	Joback Method
dvisc	0.0019096	Paxs	381.71	Joback Method
dvisc	0.0010572	Paxs	423.79	Joback Method
dvisc	0.0006513	Paxs	465.88	Joback Method
dvisc	0.0004348	Paxs	507.96	Joback Method
dvisc	0.0003088	Paxs	550.04	Joback Method
dvisc	0.0002302	Paxs	592.13	Joback Method
dvisc	0.0001784	Paxs	634.21	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	454.70	K	6.70	NIST Webbook
tbrp	414.00 ± 1.00	K	2.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1115306&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1115306&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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