

# Cyclohexane, 1-ethenyl-1-methyl-2,4-bis(1-methylethenyl)-, (1«alpha»,2«beta»,4«beta»)-

Other names: «beta»-clemene  
(1«alpha»,2«beta»,4«beta»)-1-methyl-2,4-bis(methylvinyl)-1-vinylcyclohexane  
(1«alpha»,2«beta»,4«beta»)-1-ethenyl-1-methyl-2,4-bis(1-methylethenyl)cyclohexane

**Inchi:** InChI=1S/C15H24/c1-7-15(6)9-8-13(11(2)3)10-14(15)12(4)5/h7,13-14H,1-2,4,8-10H2,3,5

**InchiKey:** OPFTUNCRGUEPRZ-UHFFFAOYSA-N

**Formula:** C15H24

**SMILES:** C=CC1(C)CCC(C(=C)C)CC1C(=C)C

**Mol. weight [g/mol]:** 204.35

**CAS:** 33880-83-0

## Physical Properties

Property code	Value	Unit	Source
gf	325.38	kJ/mol	Joback Method
hf	32.66	kJ/mol	Joback Method
hfus	15.82	kJ/mol	Joback Method
hvap	45.79	kJ/mol	Joback Method
log10ws	-4.83		Crippen Method
logp	4.747		Crippen Method
mcvol	198.450	ml/mol	McGowan Method
pc	1829.41	kPa	Joback Method
tb	542.85	K	Joback Method
tc	754.13	K	Joback Method
tf	248.41	K	Joback Method
vc	0.750	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	488.78	J/molxK	542.85	Joback Method
cpg	511.42	J/molxK	578.06	Joback Method
cpg	532.71	J/molxK	613.28	Joback Method
cpg	552.77	J/molxK	648.49	Joback Method
cpg	571.73	J/molxK	683.70	Joback Method
cpg	589.70	J/molxK	718.92	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C33880830&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C33880830&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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