

# 5-Hexyn-1-ol

<b>Other names:</b>	1-Hydroxy-5-hexyne 6-Hydroxy-1-hexyne Hex-5-yn-1-ol
<b>Inchi:</b>	InChI=1S/C6H10O/c1-2-3-4-5-6-7/h1,7H,3-6H2
<b>InchiKey:</b>	GOQJMMHTSOQIEI-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O
<b>SMILES:</b>	C#CCCCCCO
<b>Mol. weight [g/mol]:</b>	98.14
<b>CAS:</b>	928-90-5

## Physical Properties

Property code	Value	Unit	Source
gf	85.89	kJ/mol	Joback Method
hf	-27.50	kJ/mol	Joback Method
hfus	18.36	kJ/mol	Joback Method
hvap	45.49	kJ/mol	Joback Method
log10ws	-1.39		Crippen Method
logp	0.782		Crippen Method
mcvol	92.670	ml/mol	McGowan Method
pc	4140.93	kPa	Joback Method
tb	418.98	K	Joback Method
tc	590.81	K	Joback Method
tf	265.17	K	Joback Method
vc	0.352	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	180.58	J/mol×K	418.98	Joback Method
cpg	188.56	J/mol×K	447.62	Joback Method
cpg	196.19	J/mol×K	476.26	Joback Method
cpg	203.50	J/mol×K	504.90	Joback Method
cpg	210.48	J/mol×K	533.54	Joback Method
cpg	217.15	J/mol×K	562.18	Joback Method

cpg

223.54

J/mol×K

590.81

Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	347.20	K	2.00	NIST Webbook
tbrp	348.00	K	2.10	NIST Webbook
tbrp	348.20	K	2.10	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.64464e+01
Coeff. B	-4.34111e+03
Coeff. C	-5.98080e+01
Temperature range (K), min.	328.15
Temperature range (K), max.	450.15

## Sources

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C928905&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

## Legend

cpg: Ideal gas heat capacity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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