

# 3-Buten-2-one, 4-(2-methoxyphenyl)-

<b>Other names:</b>	4-(2-Methoxyphenyl)but-3-en-2-one
<b>Inchi:</b>	InChI=1S/C11H12O2/c1-9(12)7-8-10-5-3-4-6-11(10)13-2/h3-8H,1-2H3/b8-7+
<b>InchiKey:</b>	BTSZEPMWUWYLCB-BQYQJAHWSA-N
<b>Formula:</b>	C11H12O2
<b>SMILES:</b>	<chem>COc1ccccc1C=CC(C)=O</chem>
<b>Mol. weight [g/mol]:</b>	176.21
<b>CAS:</b>	10542-87-7

## Physical Properties

Property code	Value	Unit	Source
gf	-9.18	kJ/mol	Joback Method
hf	-172.89	kJ/mol	Joback Method
hfus	20.89	kJ/mol	Joback Method
hvap	52.13	kJ/mol	Joback Method
ie	8.20 ± 0.10	eV	NIST Webbook
ie	8.20	eV	NIST Webbook
log10ws	-2.53		Crippen Method
logp	2.297		Crippen Method
mcvol	145.230	ml/mol	McGowan Method
pc	2890.51	kPa	Joback Method
tb	563.19	K	Joback Method
tc	783.00	K	Joback Method
tf	319.75	K	Joback Method
vc	0.547	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	326.84	J/molxK	563.19	Joback Method
cpg	340.44	J/molxK	599.82	Joback Method
cpg	353.23	J/molxK	636.46	Joback Method
cpg	365.23	J/molxK	673.09	Joback Method
cpg	376.46	J/molxK	709.73	Joback Method
cpg	386.97	J/molxK	746.36	Joback Method

cpg	396.77	J/mol×K	783.00	Joback Method
dvisc	0.0015957	Paxs	319.75	Joback Method
dvisc	0.0008853	Paxs	360.32	Joback Method
dvisc	0.0005534	Paxs	400.90	Joback Method
dvisc	0.0003771	Paxs	441.47	Joback Method
dvisc	0.0002741	Paxs	482.04	Joback Method
dvisc	0.0002094	Paxs	522.62	Joback Method
dvisc	0.0001663	Paxs	563.19	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10542877&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10542877&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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