

# Spiro[4.5]decane, 7-hexadecyl-

<b>Other names:</b>	7-Hexadecylspiro[4.5]decane 7-n-Hexadecylspiro[4.5]decane
<b>Inchi:</b>	InChI=1S/C26H50/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-19-25-20-18-23-26(24-25)21-1
<b>InchiKey:</b>	BNDDTWKJQMCKBA-UHFFFAOYSA-N
<b>Formula:</b>	C26H50
<b>SMILES:</b>	CCCCCCCCCCCCCCCC1CCCC2(CCCC2)C1
<b>Mol. weight [g/mol]:</b>	362.68
<b>CAS:</b>	2307-06-4

## Physical Properties

Property code	Value	Unit	Source
chl	-16786.50 ± 2.20	kJ/mol	NIST Webbook
gf	235.65	kJ/mol	Joback Method
hf	-443.77	kJ/mol	Joback Method
hfl	-590.40 ± 2.40	kJ/mol	NIST Webbook
hfus	44.67	kJ/mol	Joback Method
hvap	72.83	kJ/mol	Joback Method
log10ws	-10.01		Crippen Method
logp	9.608		Crippen Method
mcvol	355.480	ml/mol	McGowan Method
pc	913.29	kPa	Joback Method
tb	825.08	K	Joback Method
tc	1019.91	K	Joback Method
tf	428.48	K	Joback Method
vc	1.371	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1204.12	J/mol×K	825.08	Joback Method
cpg	1230.04	J/mol×K	857.55	Joback Method
cpg	1255.03	J/mol×K	890.02	Joback Method
cpg	1279.23	J/mol×K	922.49	Joback Method
cpg	1302.78	J/mol×K	954.97	Joback Method

cpg	1325.81	J/mol×K	987.44	Joback Method
cpg	1348.47	J/mol×K	1019.91	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.19832e+01
Coeff. B	-5.17298e+03
Temperature range (K), min.	442.30
Temperature range (K), max.	775.36

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2307064&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2307064&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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