

# «alpha»,«alpha»,«alpha»-Trifluoro-o-tolunitrile

<b>Other names:</b>	o-Trifluoromethylbenzotrile 2-Trifluoromethylbenzotrile Benzotrile, 2-(trifluoromethyl)- o-Tolunitrile, alpha,alpha,alpha-trifluoro- alpha,alpha,alpha-Trifluoro-o-tolunitrile «alpha», «alpha», «alpha»-trifluoro-2-toluonitrile
<b>Inchi:</b>	InChI=1S/C8H4F3N/c9-8(10,11)7-4-2-1-3-6(7)5-12/h1-4H
<b>InchiKey:</b>	SOZGHDCEWOLLHV-UHFFFAOYSA-N
<b>Formula:</b>	C8H4F3N
<b>SMILES:</b>	N#Cc1cccc1C(F)(F)F
<b>Mol. weight [g/mol]:</b>	171.12
<b>CAS:</b>	447-60-9

## Physical Properties

Property code	Value	Unit	Source
ea	0.70 ± 0.10	eV	NIST Webbook
gf	-329.15	kJ/mol	Joback Method
hf	-415.59	kJ/mol	Joback Method
hfus	13.46	kJ/mol	Joback Method
hvap	43.07	kJ/mol	Joback Method
log10ws	-2.93		Crippen Method
logp	2.577		Crippen Method
mvol	106.510	ml/mol	McGowan Method
pc	3049.04	kPa	Joback Method
tb	478.00 ± 1.00	K	NIST Webbook
tc	722.52	K	Joback Method
tf	288.04	K	Joback Method
vc	0.445	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	229.98	J/mol×K	510.76	Joback Method
cpg	239.29	J/mol×K	546.05	Joback Method

cpg	247.87	J/mol×K	581.35	Joback Method
cpg	255.76	J/mol×K	616.64	Joback Method
cpg	263.01	J/mol×K	651.93	Joback Method
cpg	269.67	J/mol×K	687.23	Joback Method
cpg	275.78	J/mol×K	722.52	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C447609&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C447609&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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