

# Racemic-2,3-dimethyl-butanedioic acid

<b>Other names:</b>	Racemic-2,3-dimethyl-succinic acid Butanedioic acid, 2,3-dimethyl-, (R*,S*)-erythro-2,3-Dimethylsuccinic acid Butanedioic acid, 2,3-dimethyl-, (R-(R*,S*))-Succinic acid, 2,3-dimethyl-, erythro-meso-Succinic acid, 2,3-dimethyl-, meso-(R*,S*)-2,3-dimethylsuccinic acid
<b>Inchi:</b>	InChI=1S/C6H10O4/c1-3(5(7)8)4(2)6(9)10/h3-4H,1-2H3,(H,7,8)(H,9,10)
<b>InchiKey:</b>	KLZYRCVPDWTZLH-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O4
<b>SMILES:</b>	CC(C(=O)O)C(C)C(=O)O
<b>Mol. weight [g/mol]:</b>	146.14
<b>CAS:</b>	608-40-2

## Physical Properties

Property code	Value	Unit	Source
chs	-2806.40 ± 1.40	kJ/mol	NIST Webbook
gf	-536.72	kJ/mol	Joback Method
hf	-707.35	kJ/mol	Joback Method
hfus	15.62	kJ/mol	Joback Method
hvap	75.02	kJ/mol	Joback Method
log10ws	-0.05		Crippen Method
logp	0.428		Crippen Method
mcvol	110.280	ml/mol	McGowan Method
pc	4640.32	kPa	Joback Method
tb	627.90	K	Joback Method
tc	807.19	K	Joback Method
tf	463.15 ± 3.00	K	NIST Webbook
vc	0.409	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.61	J/mol×K	627.90	Joback Method

cpg	311.60	J/molxK	777.31	Joback Method
cpg	305.72	J/molxK	747.43	Joback Method
cpg	299.49	J/molxK	717.55	Joback Method
cpg	292.90	J/molxK	687.66	Joback Method
cpg	285.95	J/molxK	657.78	Joback Method
cpg	317.14	J/molxK	807.19	Joback Method
dvisc	0.0000330	Paxs	627.90	Joback Method
dvisc	0.0000614	Paxs	581.40	Joback Method
dvisc	0.0001275	Paxs	534.89	Joback Method
dvisc	0.0003041	Paxs	488.39	Joback Method
dvisc	0.0008709	Paxs	441.89	Joback Method
dvisc	0.0031943	Paxs	395.38	Joback Method
dvisc	0.0165669	Paxs	348.88	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C608402&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C608402&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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