

# iso-3-Thujyl acetate

**Other names:**

Isothujyl acetate  
Neo-3-Thujyl alcohol, acetate  
Iso-3-Thujyl alcohol, acetate  
isothujyl-3-acetate  
Thujyl acetate(3-iso)

**Inchi:**

InChI=1S/C12H20O2/c1-7(2)12-5-10(12)8(3)11(6-12)14-9(4)13/h7-8,10-11H,5-6H2,1-4H

**InchiKey:**

RYMWIDNPMDLHRP-UHFFFAOYSA-N

**Formula:**

C12H20O2

**SMILES:**

CC(=O)OC1CC2(C(C)C)CC2C1C

**Mol. weight [g/mol]:**

196.29

**CAS:**

62181-90-2

## Physical Properties

Property code	Value	Unit	Source
gf	-85.61	kJ/mol	Joback Method
hf	-420.93	kJ/mol	Joback Method
hfus	18.21	kJ/mol	Joback Method
hvap	49.13	kJ/mol	Joback Method
log10ws	-2.64		Crippen Method
logp	2.620		Crippen Method
mcvol	165.660	ml/mol	McGowan Method
pc	2318.07	kPa	Joback Method
rinpol	1301.00		NIST Webbook
rinpol	1273.00		NIST Webbook
rinpol	1298.00		NIST Webbook
rinpol	1305.00		NIST Webbook
rinpol	1270.00		NIST Webbook
rinpol	1301.00		NIST Webbook
rinpol	1279.00		NIST Webbook
rinpol	1291.00		NIST Webbook
rinpol	1266.50		NIST Webbook
rinpol	1291.00		NIST Webbook
rinpol	1261.00		NIST Webbook
rinpol	1270.00		NIST Webbook
rinpol	1291.00		NIST Webbook
rinpol	1294.00		NIST Webbook
rinpol	1266.50		NIST Webbook

rinpol	1301.00		NIST Webbook
rinpol	1297.00		NIST Webbook
rinpol	1294.00		NIST Webbook
rinpol	1291.00		NIST Webbook
ripol	1630.00		NIST Webbook
tb	554.19	K	Joback Method
tc	758.84	K	Joback Method
tf	333.46	K	Joback Method
vc	0.635	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	438.10	J/mol×K	554.19	Joback Method
cpg	456.42	J/mol×K	588.30	Joback Method
cpg	473.65	J/mol×K	622.41	Joback Method
cpg	489.90	J/mol×K	656.52	Joback Method
cpg	505.31	J/mol×K	690.62	Joback Method
cpg	519.99	J/mol×K	724.73	Joback Method
cpg	534.06	J/mol×K	758.84	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C62181902&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C62181902&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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