

# Cyclopentanol,2-methoxy-,trans-

<b>Inchi:</b>	InChI=1S/C6H12O2/c1-8-6-4-2-3-5(6)7/h5-7H,2-4H2,1H3/t5-,6-/m0/s1
<b>InchiKey:</b>	YVXLDOASGKIXII-WDSKDSINSA-N
<b>Formula:</b>	C6H12O2
<b>SMILES:</b>	COC1CCCC1O
<b>Mol. weight [g/mol]:</b>	116.16
<b>CAS:</b>	7429-45-0

## Physical Properties

Property code	Value	Unit	Source
gf	-213.34	kJ/mol	Joback Method
hf	-411.48	kJ/mol	Joback Method
hfus	11.58	kJ/mol	Joback Method
hvap	47.99	kJ/mol	Joback Method
ie	9.60	eV	NIST Webbook
log10ws	-0.80		Crippen Method
logp	0.546		Crippen Method
mcvol	96.280	ml/mol	McGowan Method
pc	4046.64	kPa	Joback Method
tb	461.89	K	Joback Method
tc	648.35	K	Joback Method
tf	247.09	K	Joback Method
vc	0.348	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	213.71	J/molxK	461.89	Joback Method
cpg	225.75	J/molxK	492.97	Joback Method
cpg	237.29	J/molxK	524.04	Joback Method
cpg	248.34	J/molxK	555.12	Joback Method
cpg	258.91	J/molxK	586.19	Joback Method
cpg	269.01	J/molxK	617.27	Joback Method
cpg	278.63	J/molxK	648.35	Joback Method
dvisc	0.0209056	Paxs	247.09	Joback Method

dvisc	0.0061775	Paxs	282.89	Joback Method
dvisc	0.0024006	Paxs	318.69	Joback Method
dvisc	0.0011291	Paxs	354.49	Joback Method
dvisc	0.0006099	Paxs	390.29	Joback Method
dvisc	0.0003653	Paxs	426.09	Joback Method
dvisc	0.0002369	Paxs	461.89	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7429450&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7429450&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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