

# 1,2,3,4-Tetrahydro-1-hydroperoxynaphthalene

<b>Inchi:</b>	InChI=1S/C10H12O2/c11-12-10-7-3-5-8-4-1-2-6-9(8)10/h1-2,4,6,10-11H,3,5,7H2
<b>InchiKey:</b>	YWBMNCRJFZGXJY-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O2
<b>SMILES:</b>	OOC1CCCc2ccccc21
<b>Mol. weight [g/mol]:</b>	164.20
<b>CAS:</b>	26447-24-5

## Physical Properties

Property code	Value	Unit	Source
chs	-5468.50 ± 8.40	kJ/mol	NIST Webbook
chs	-5462.60	kJ/mol	NIST Webbook
gf	-57.07	kJ/mol	Joback Method
hf	-242.48	kJ/mol	Joback Method
hfs	-187.40 ± 5.40	kJ/mol	NIST Webbook
hfus	16.62	kJ/mol	Joback Method
hvap	59.97	kJ/mol	Joback Method
log10ws	-2.83		Crippen Method
logp	2.554		Crippen Method
mcvol	128.880	ml/mol	McGowan Method
pc	3754.57	kPa	Joback Method
tb	585.47	K	Joback Method
tc	797.79	K	Joback Method
tf	338.87	K	Joback Method
vc	0.473	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	321.32	J/molxK	585.47	Joback Method
cpg	334.62	J/molxK	620.86	Joback Method
cpg	347.09	J/molxK	656.24	Joback Method
cpg	358.77	J/molxK	691.63	Joback Method
cpg	369.69	J/molxK	727.02	Joback Method
cpg	379.89	J/molxK	762.41	Joback Method

cpg	389.40	J/mol×K	797.79	Joback Method
dvisc	0.0041419	Paxs	338.87	Joback Method
dvisc	0.0017090	Paxs	379.97	Joback Method
dvisc	0.0008382	Paxs	421.07	Joback Method
dvisc	0.0004666	Paxs	462.17	Joback Method
dvisc	0.0002858	Paxs	503.27	Joback Method
dvisc	0.0001885	Paxs	544.37	Joback Method
dvisc	0.0001319	Paxs	585.47	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C26447245&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C26447245&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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