

Ethane, isocyanato-

Other names:	C2H5NCO Ethyl isocyanate Isocyanatoethane Isocyanic acid, ethyl ester UN 2481
Inchi:	InChI=1S/C3H5NO/c1-2-4-3-5/h2H2,1H3
InchiKey:	WUDNUHPRLBTKOJ-UHFFFAOYSA-N
Formula:	C3H5NO
SMILES:	CCN=C=O
Mol. weight [g/mol]:	71.08
CAS:	109-90-0

Physical Properties

Property code	Value	Unit	Source
hf	-110.66	kJ/mol	Joback Method
hvap	31.80	kJ/mol	Joback Method
ie	10.10	eV	NIST Webbook
ie	10.32 ± 0.05	eV	NIST Webbook
log10ws	-4.58		Crippen Method
logp	0.342		Crippen Method
mcvol	60.380	ml/mol	McGowan Method
pc	4876.56	kPa	Joback Method
tb	334.71	K	Joback Method
tc	514.09	K	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52808e+01
Coeff. B	-3.23484e+03
Coeff. C	-3.37660e+01
Temperature range (K), min.	249.52

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C109900&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature

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