

Diethylmalonic acid, 2-fluoroethyl tridecyl ester

Inchi:	InChI=1S/C22H41FO4/c1-4-7-8-9-10-11-12-13-14-15-16-18-26-20(24)22(5-2,6-3)21(25)2
InchiKey:	IMLQSXHGGGBCNR-UHFFFAOYSA-N
Formula:	C22H41FO4
SMILES:	CCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCCF
Mol. weight [g/mol]:	388.56

Physical Properties

Property code	Value	Unit	Source
gf	-525.45	kJ/mol	Joback Method
hf	-1191.87	kJ/mol	Joback Method
hfus	53.98	kJ/mol	Joback Method
hvap	80.77	kJ/mol	Joback Method
log10ws	-6.37		Crippen Method
logp	6.160		Crippen Method
mcvol	337.490	ml/mol	McGowan Method
pc	938.64	kPa	Joback Method
rinsol	2361.00		NIST Webbook
tb	851.38	K	Joback Method
tc	1042.47	K	Joback Method
tf	485.03	K	Joback Method
vc	1.323	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1099.80	J/mol×K	851.38	Joback Method
cpg	1118.60	J/mol×K	883.23	Joback Method
cpg	1136.24	J/mol×K	915.08	Joback Method
cpg	1152.76	J/mol×K	946.92	Joback Method
cpg	1168.19	J/mol×K	978.77	Joback Method
cpg	1182.59	J/mol×K	1010.62	Joback Method
cpg	1195.99	J/mol×K	1042.47	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U370867&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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