

# Diethylmalonic acid, 2-formylphenyl pentadecyl ester

**Inchi:** InChI=1S/C29H46O5/c1-4-7-8-9-10-11-12-13-14-15-16-17-20-23-33-27(31)29(5-2,6-3)28

**InchiKey:** VBMYQRNKYXQACO-UHFFFAOYSA-N

**Formula:** C29H46O5

**SMILES:** CCCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccccc1C=O

**Mol. weight [g/mol]:** 474.67

## Physical Properties

| Property code | Value    | Unit    | Source         |
|---------------|----------|---------|----------------|
| gf            | -268.44  | kJ/mol  | Joback Method  |
| hf            | -1000.76 | kJ/mol  | Joback Method  |
| hfus          | 64.97    | kJ/mol  | Joback Method  |
| hvap          | 106.82   | kJ/mol  | Joback Method  |
| log10ws       | -9.02    |         | Crippen Method |
| logp          | 7.845    |         | Crippen Method |
| mcvol         | 412.160  | ml/mol  | McGowan Method |
| pc            | 815.86   | kPa     | Joback Method  |
| rinpol        | 3277.00  |         | NIST Webbook   |
| rinpol        | 3277.00  |         | NIST Webbook   |
| tb            | 1092.59  | K       | Joback Method  |
| tc            | 1347.62  | K       | Joback Method  |
| tf            | 644.27   | K       | Joback Method  |
| vc            | 1.605    | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 1444.60   | J/mol×K | 1092.59         | Joback Method |
| cpg           | 1461.49   | J/mol×K | 1135.10         | Joback Method |
| cpg           | 1476.63   | J/mol×K | 1177.60         | Joback Method |
| cpg           | 1490.17   | J/mol×K | 1220.11         | Joback Method |
| cpg           | 1502.22   | J/mol×K | 1262.61         | Joback Method |
| cpg           | 1512.91   | J/mol×K | 1305.12         | Joback Method |
| cpg           | 1522.37   | J/mol×K | 1347.62         | Joback Method |
| dvisc         | 0.0001610 | Pa×s    | 644.27          | Joback Method |

|       |           |      |         |               |
|-------|-----------|------|---------|---------------|
| dvisc | 0.0000819 | Pa×s | 718.99  | Joback Method |
| dvisc | 0.0000473 | Pa×s | 793.71  | Joback Method |
| dvisc | 0.0000300 | Pa×s | 868.43  | Joback Method |
| dvisc | 0.0000205 | Pa×s | 943.15  | Joback Method |
| dvisc | 0.0000148 | Pa×s | 1017.87 | Joback Method |
| dvisc | 0.0000111 | Pa×s | 1092.59 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369960&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369960&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |

## Legend

|                            |                                                 |
|----------------------------|-------------------------------------------------|
| <b>cp<sub>g</sub>:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>              | Dynamic viscosity                               |
| <b>g<sub>f</sub>:</b>      | Standard Gibbs free energy of formation         |
| <b>h<sub>f</sub>:</b>      | Enthalpy of formation at standard conditions    |
| <b>h<sub>fus</sub>:</b>    | Enthalpy of fusion at standard conditions       |
| <b>h<sub>vap</sub>:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log<sub>10ws</sub>:</b> | Log10 of Water solubility in mol/l              |
| <b>log<sub>p</sub>:</b>    | Octanol/Water partition coefficient             |
| <b>mc<sub>vol</sub>:</b>   | McGowan's characteristic volume                 |
| <b>p<sub>c</sub>:</b>      | Critical Pressure                               |
| <b>rin<sub>pol</sub>:</b>  | Non-polar retention indices                     |
| <b>t<sub>b</sub>:</b>      | Normal Boiling Point Temperature                |
| <b>t<sub>c</sub>:</b>      | Critical Temperature                            |
| <b>t<sub>f</sub>:</b>      | Normal melting (fusion) point                   |
| <b>v<sub>c</sub>:</b>      | Critical Volume                                 |

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