

# Isophthalic acid, dodecyl pent-4-enyl ester

<b>Inchi:</b>	InChI=1S/C25H38O4/c1-3-5-7-8-9-10-11-12-13-15-20-29-25(27)23-18-16-17-22(21-23)2
<b>InchiKey:</b>	HFMXJEPQOVJJQJ-UHFFFAOYSA-N
<b>Formula:</b>	C25H38O4
<b>SMILES:</b>	C=CCCCOC(=O)c1cccc(C(=O)OCCCCCCCCCCCC)c1
<b>Mol. weight [g/mol]:</b>	402.57

## Physical Properties

Property code	Value	Unit	Source
gf	-117.60	kJ/mol	Joback Method
hf	-698.44	kJ/mol	Joback Method
hfus	58.45	kJ/mol	Joback Method
hvap	91.82	kJ/mol	Joback Method
log10ws	-8.09		Crippen Method
logp	6.887		Crippen Method
mcvol	349.930	ml/mol	McGowan Method
pc	996.39	kPa	Joback Method
rinpol	2950.00		NIST Webbook
tb	952.32	K	Joback Method
tc	1166.01	K	Joback Method
tf	553.01	K	Joback Method
vc	1.357	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1151.84	J/molxK	952.32	Joback Method
cpg	1222.77	J/molxK	1130.40	Joback Method
cpg	1211.14	J/molxK	1094.78	Joback Method
cpg	1198.27	J/molxK	1059.17	Joback Method
cpg	1184.14	J/molxK	1023.55	Joback Method
cpg	1168.68	J/molxK	987.94	Joback Method
cpg	1233.23	J/molxK	1166.01	Joback Method
dvisc	0.0000286	Paxs	952.32	Joback Method
dvisc	0.0000372	Paxs	885.77	Joback Method

dvisc	0.0000504	Paxs	819.22	Joback Method
dvisc	0.0000721	Paxs	752.66	Joback Method
dvisc	0.0001105	Paxs	686.11	Joback Method
dvisc	0.0001858	Paxs	619.56	Joback Method
dvisc	0.0003540	Paxs	553.01	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U356722&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U356722&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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