

# Hexane, 3-ethyl-4-methyl-

<b>Other names:</b>	3-ETHYL-4-METHYLHEXANE 3-METHYL-4-ETHYLHEXANE 4-ETHYL-3-METHYLHEXANE Hexane, 4-ethyl-3-methyl
<b>Inchi:</b>	InChI=1S/C9H20/c1-5-8(4)9(6-2)7-3/h8-9H,5-7H2,1-4H3
<b>InchiKey:</b>	OKCRKWVABWILDR-UHFFFAOYSA-N
<b>Formula:</b>	C9H20
<b>SMILES:</b>	CCC(C)C(CC)CC
<b>Mol. weight [g/mol]:</b>	128.26
<b>CAS:</b>	3074-77-9

## Physical Properties

Property code	Value	Unit	Source
af	0.3720		KDB
gf	20.02	kJ/mol	Joback Method
hcg	6124.08	kJ/mol	KDB
hcn	5683.964	kJ/mol	KDB
hf	-239.65	kJ/mol	Joback Method
hfus	12.02	kJ/mol	Joback Method
hvap	34.85	kJ/mol	Joback Method
log10ws	-3.11		Crippen Method
logp	3.469		Crippen Method
mcvol	137.670	ml/mol	McGowan Method
pc	2510.00	kPa	KDB
rinpol	856.00		NIST Webbook
rinpol	856.00		NIST Webbook
rinpol	851.00		NIST Webbook
rinpol	853.60		NIST Webbook
rinpol	856.00		NIST Webbook
rinpol	855.60		NIST Webbook
rinpol	856.00		NIST Webbook
rinpol	856.30		NIST Webbook
rinpol	856.00		NIST Webbook
rinpol	856.00		NIST Webbook
rinpol	860.00		NIST Webbook
rinpol	856.00		NIST Webbook
rinpol	857.10		NIST Webbook

rinpol	856.00		NIST Webbook
tb	413.55 ± 0.50	K	NIST Webbook
tb	413.60	K	KDB
tc	593.70	K	KDB
tf	160.00	K	KDB
vc	0.490	m <sup>3</sup> /kmol	KDB
zc	0.2491530		KDB

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	272.52	J/mol×K	404.44	Joback Method
cpg	287.44	J/mol×K	432.85	Joback Method
cpg	301.80	J/mol×K	461.26	Joback Method
cpg	315.61	J/mol×K	489.67	Joback Method
cpg	328.88	J/mol×K	518.07	Joback Method
cpg	341.62	J/mol×K	546.48	Joback Method
cpg	353.86	J/mol×K	574.89	Joback Method
dvisc	0.0048330	Paxs	201.73	Joback Method
dvisc	0.0224972	Paxs	161.19	Joback Method
dvisc	0.0017372	Paxs	242.27	Joback Method
dvisc	0.0008373	Paxs	282.81	Joback Method
dvisc	0.0004846	Paxs	323.36	Joback Method
dvisc	0.0003168	Paxs	363.90	Joback Method
dvisc	0.0002255	Paxs	404.44	Joback Method
hvapt	36.40	kJ/mol	413.60	KDB
rfi	1.41280		298.15	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42302e+01
Coeff. B	-3.43058e+03
Coeff. C	-5.66490e+01
Temperature range (K), min.	302.70
Temperature range (K), max.	441.30

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.92974e+01
Coeff. B	-8.08141e+03
Coeff. C	-1.10102e+01
Coeff. D	7.01203e-06
Temperature range (K), min.	302.15
Temperature range (K), max.	593.70

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=81">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=81</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3074779&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3074779&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=81">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=81</a>

## Legend

<b>af:</b>	Acentric Factor
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hcg:</b>	Heat of Combustion, Gross form
<b>hcn:</b>	Heat of Combustion, Net Form
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rfi:</b>	Refractive Index
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume
<b>zc:</b>	Critical Compressibility

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