

# Octanoic acid, phenylmethyl ester

<b>Other names:</b>	Octanoic acid, benzyl ester Benzyl n-octanoate Benzyl octanoate Benzyl n-octanate Benzyl caprylate
<b>Inchi:</b>	InChI=1S/C15H22O2/c1-2-3-4-5-9-12-15(16)17-13-14-10-7-6-8-11-14/h6-8,10-11H,2-5,9
<b>InchiKey:</b>	MWQWCHLIPMDVLS-UHFFFAOYSA-N
<b>Formula:</b>	C15H22O2
<b>SMILES:</b>	CCCCCCCC(=O)OCc1ccccc1
<b>Mol. weight [g/mol]:</b>	234.33
<b>CAS:</b>	10276-85-4

## Physical Properties

Property code	Value	Unit	Source
gf	-46.09	kJ/mol	Joback Method
hf	-361.20	kJ/mol	Joback Method
hfus	31.43	kJ/mol	Joback Method
hvap	60.42	kJ/mol	Joback Method
log10ws	-4.56		Crippen Method
logp	4.090		Crippen Method
mcvol	205.890	ml/mol	McGowan Method
pc	1921.98	kPa	Joback Method
rinpol	1720.00		NIST Webbook
rinpol	1726.00		NIST Webbook
rinpol	1707.00		NIST Webbook
ripol	2260.00		NIST Webbook
tb	645.57	K	Joback Method
tc	843.08	K	Joback Method
tf	357.39	K	Joback Method
vc	0.791	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	546.39	J/molxK	645.57	Joback Method
cpg	563.17	J/molxK	678.49	Joback Method
cpg	579.02	J/molxK	711.41	Joback Method
cpg	593.96	J/molxK	744.32	Joback Method
cpg	608.02	J/molxK	777.24	Joback Method
cpg	621.23	J/molxK	810.16	Joback Method
cpg	633.62	J/molxK	843.08	Joback Method
dvisc	0.0019370	Paxs	357.39	Joback Method
dvisc	0.0009578	Paxs	405.42	Joback Method
dvisc	0.0005498	Paxs	453.45	Joback Method
dvisc	0.0003510	Paxs	501.48	Joback Method
dvisc	0.0002424	Paxs	549.51	Joback Method
dvisc	0.0001776	Paxs	597.54	Joback Method
dvisc	0.0001364	Paxs	645.57	Joback Method

## Sources

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C10276854&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/67-606-5/Octanoic-acid-phenylmethyl-ester.pdf>

Generated by Cheméo on 2024-04-25 20:10:24.842104025 +0000 UTC m=+16365073.762681341.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.