

# anastreptene

<b>Other names:</b>	(+)-Anastreptene
<b>Inchi:</b>	InChI=1S/C15H22/c1-9-5-6-11-14(4)8-7-10-12(13(10,2)3)15(9,11)14/h5,10-12H,6-8H2,1
<b>InchiKey:</b>	SYNYVXGDEQOMCB-SEOXFTARSA-N
<b>Formula:</b>	C15H22
<b>SMILES:</b>	CC1=CCC2C3(C)CCC4C(C4(C)C)C123
<b>Mol. weight [g/mol]:</b>	202.34

## Physical Properties

Property code	Value	Unit	Source
gf	318.96	kJ/mol	Joback Method
hf	-4.22	kJ/mol	Joback Method
hfus	13.33	kJ/mol	Joback Method
hvap	45.35	kJ/mol	Joback Method
log10ws	-4.08		Crippen Method
logp	4.025		Crippen Method
mcvol	174.470	ml/mol	McGowan Method
pc	2327.03	kPa	Joback Method
rinpol	1391.00		NIST Webbook
rinpol	1391.00		NIST Webbook
rinpol	1370.00		NIST Webbook
rinpol	1391.00		NIST Webbook
rinpol	1403.00		NIST Webbook
rinpol	1405.00		NIST Webbook
rinpol	1400.00		NIST Webbook
rinpol	1404.00		NIST Webbook
rinpol	1404.00		NIST Webbook
rinpol	1403.60		NIST Webbook
rinpol	1403.00		NIST Webbook
ripol	1568.00		NIST Webbook
ripol	1564.00		NIST Webbook
ripol	1568.00		NIST Webbook
ripol	1568.00		NIST Webbook
ripol	1570.00		NIST Webbook
ripol	1582.00		NIST Webbook
ripol	1564.00		NIST Webbook
ripol	1564.00		NIST Webbook
ripol	1564.00		NIST Webbook

ripol	1569.00		NIST Webbook
ripol	1575.00		NIST Webbook
ripol	1577.00		NIST Webbook
ripol	1579.00		NIST Webbook
tb	560.81	K	Joback Method
tc	788.13	K	Joback Method
tf	410.59	K	Joback Method
vc	0.690	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	484.78	J/mol×K	560.81	Joback Method
cpg	505.53	J/mol×K	598.70	Joback Method
cpg	524.57	J/mol×K	636.58	Joback Method
cpg	542.38	J/mol×K	674.47	Joback Method
cpg	559.40	J/mol×K	712.36	Joback Method
cpg	576.11	J/mol×K	750.24	Joback Method
cpg	592.97	J/mol×K	788.13	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R141830&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R141830&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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